## **Amide Bond Replacements Incorporated into CCK-B Selective "Dipeptoids"**

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This paper describes the chemical synthesis and CCK-B and CCK-A receptor binding affinities of a series of compounds in which the central amide bond of the CCK-B "dipeptoid" ligand tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yl  $[R-(R^*,S^*)]$ -[2-[[1-(hydroxymethyl)-2-phenylethyl]amino]-1-(1H-indol-3-ylmethyl)-2-oxoethyl]carbamate (4) (CCK-B IC<sub>50</sub> = 852 nM), and tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yl (*R*)-[1-(1*H*-indol-3-ylmethyl)-1-methyl-2-oxo-2-[(2-phenylethyl)amino]ethyl]carbamate (23) (CCK-B IC<sub>50</sub> = 32 nM) is replaced by 11 different amide replacements. These replacements are the methyleneamino  $(CH_2NH)$ , the reverse amide (NHCO), the ester (COO), the N-methylamide (CONMe), the thioamide (CSNH), the  $N$ -acetylmethyleneamino (CH<sub>2</sub>NAc), the cis double bond (CHCH), the ethylene (CH<sub>2</sub>CH<sub>2</sub>), the thiolester (COS), the hydroxyethylene (CHOHCH2), and a 4,5-dihydro-l,3-thiazole. Most of the replacements have weaker affinity and reduced selectivity for the CCK-B receptor than the parent amide. However, this affinity can be improved by appending a fumarate side chain to the phenethyl group, e.g. tricyclo $[3.3.1.1^{3.7}]$ dec-2-yl-3- $(1H$ -indol-3-ylmethyl)-3-methyl-4,9-dioxo-7-phenyl-5,13-dioxa-2,8-diazatetradec-10-enoate (36) (CCK-B IC<sub>50</sub> = 38.8 nM). Replacement of the amide of compound 4 with a 4,5-dihydro-l,3-thiazole gives tricyclo[3.3.1.13>7]dec-2-yl [l-[4,5-dihydro-4- (phenylmethyl)-2-thiazolyl]-2-(LH-indol-3-yl)ethyl]carbamate (5), which is selective for the CCK-A receptor (CCK-A  $IC_{50} = 125$  nM, CCK-B IC<sub>50</sub> = 2580 nM, ratio = 21). The methyleneamino and hydroxyethylene replacements, which have been used elsewhere as transition-state inhibitors of enzymes, are poor mimics of the amide in these CCK-B receptor ligands. Some of the steric, lipophilic, and hydrogen bonding properties of amide replacements incorporated into the simple amide, N-methylacetamide, have been quantified with the aid of molecular modeling. These data will contribute to the rational selection of amide bond replacements in other substrates.

#### **Introduction**

The amide bonds of peptides constitute a major site of enzymatic metabolism and limit the use of peptides as drugs.<sup>1</sup> This has led to a large amount of work aimed at replacing the amide bond (CONH) of biologically active peptides with a variety of groups. These include a reversed amide (NHCO), a hydroxyethylene  $(CHOHCH<sub>2</sub>)$ , a methyleneamino (NHCH<sub>2</sub>), a ketomethylene (COCH<sub>2</sub>), or an olefin (HC=CH) in an attempt to identify metabolically stable agonists/antagonists of peptide receptors or proteolytic enzyme inhibitors. These backbone modifications have been reviewed by Spatola<sup>2</sup> and by the Royal Society of Chemistry annually since 1984.<sup>3</sup> These reviews describe modifications (frequently referred to by the term "isosteres") incorporated into one or more of the amide bonds of several neuropeptides and hormones including Met- and Leu-enkephalin, cholecystokinin (CCK), gastrin, and somatostatin in an attempt to mimic the receptorbound state of the amide. These modifications have also been used extensively in enzyme inhibitors [e.g. the metalloprotease, angiotensin converting enzyme (EC 3.4.15.11), and the aspartic proteinases, renin (EC 3.4.99.19) and HIV proteinase] to mimic an enzyme-bound  $\alpha$ .  $\alpha$ .  $\beta$ .  $\beta$  and  $\pi$  iv procedure is a minimized enzyme-bound. terest in this area continues to grow with several publiterest in this area continues to grow with several publi-<br>cations describing new types of amide replacements<sup>6</sup> or the chemical synthesis of classical ground or transition state "amide isosteres" into peptide derivatives.<sup>6</sup>

- (2) Spatola, A. F. Peptide Backbone Modifications: A Structure-Activity Analysis of Peptides containing Amide Bond Surrogates. In *Chemistry and Biochemistry of Amino Acids, Peptides and Proteins;* Weinstein, B., Ed.; Dekker: New York, 1983; Vol. 7, Chapter 5.
- (3) (a) Davies, J. S. Analogue and Conformational Studies on Peptide Hormones and Other Biologically Active Peptides. In *Amino Acids and Peptides;* Jones, J. H., Ed.; The Royal Society of Chemistry: London, 1986; Vol. 17, pp 104-125. (b) Ibid. 1987; Vol. 18, pp 140-168. (c) Ibid. 1987; Vol. 19, pp 141-189. (d) Ibid. 1989; Vol. 20, pp 128-174. (e) Ibid. 1990; Vol. 21, pp 129-173.

As part of our program to design non-peptide ligands ("peptoids") which modulate the actions of mammalian peptide neurotransmitters or hormones, we have recently described the discovery of a novel series of selective "dipeptoid" cholecystokinin-B (CCK-B) and gastrin receptor antagonists of which the prototype is CI-988 (PD  $134308$ )  $(1)$ .<sup>7</sup> These compounds, which were developed from the neuropeptide CCK-26-33 (sulfated) (2) have been shown to have potent anxiolytic and anti-gastrin activity in rodent models.<sup>7</sup>



**(2)** 

**CCK 26-33 Asp-Tyr-(S03H)-Met-Gly-Trp-Met-Asp-Phe-NH<sup>2</sup>**

- (4) (a) Rich, D. H. Peptidase Inhibitors. In *Comprehensive Medicinal Chemistry;* Hansen, C, Sammes, P. G., Taylor, J. B., Eds.; Pergamon: New York, 1990; Vol. 2., Chapter 8.2, pp 391-441. (b) Page, M. I. Enzyme Inhibition. Ibid.; Vol. 2, Chapter 5.4, pp 61-87. (c) Roberts, N. A.; Martin, J. A.; Kinchington, D.; Broadhurst, A. V.; Craig, J. C; Duncan, I. B.; Galpin, S. A.; Handa, B. K.; Kay, J.; Krohn, A.; Lambert, R. W.; Merrett, J. H.; Mills, J. S.; Parkes, K. E. B.; Redshaw, S.; Ritchie, A. J.; Taylor, D. L.; Thomas, G.; Machin, P. J.; Rational Design of Peptide-Based HIV Proteinase Inhibitors. *Science* 1990, *248,* 358-361.
- (5) (a) Wolf, J-P.; Rapoport, H. Conformationally Constrained Peptides. Chirospecific Synthesis of 4-Alkyl-Substituted  $\gamma$ -Lactam-Bridged Dipeptides from L-Aspartic Acid. *J. Org. Chem.* 1989, *54,* 3164-3173. (b) Jones, R. C. F.; Ward, G. J.; Amide Bond Isosteres: Imidazolines in Pseudopeptide Chemistry. *Tetrahedron Lett.* 1988,29, 3853-3856. (c) Arrowsmith, R. J.; Davies, D. E.; Fogden, Y. C; Harris, C. J.; Thompson, C; Aminc-Alcohol Dipeptide Analogues: A Simple Synthesis of a Versatile Isostere for the Development of Proteinase Inhibitors. *Tetrahedron Lett.* 1987, *28,* 5569-5572.

<sup>(1)</sup> Humphrey, M. J.; Ringrose, P. S. Peptides and Related Drugs: A Review of their Absorption, Metabolism and Excretion. *Drug Metab. Rev.* 1986,*17,* 283-310.





<sup>a</sup>(a) (i) DCC, PFP, EtOAc; (ii) (S)-2-amino-3-phenyl-1-propanol; (b) Lawesson's reagent; (c) (i) N-methylmorpholine, i-BuOCOCl; (ii)  $\text{Me}_3\text{SiN}_3$ ; (iii) 40 °C; (iv) p-nitrobenzyl alcohol, DABCO; (d) (i) 20% Pd(OH)<sub>2</sub>/C, H<sub>2</sub>, 45 psi, 40 °C; (ii) compound 21, HOBt, DCC; (iii) LiOH; (e) PFP, DCC, DMAP, (-)-(S)-3-phenyllactic acid, DMF; (f) (i) EtOCOCl, EtsN; (ii) NaBH4; (g) (i) iV-methylmorpholine, i-BuOCOCl; (ii) MeNH(OMe)-HCl; (h) LiAlH4; (i) NaCN(BH3), (S)-2-amino-3-phenyl-l-propanol; (j) (i) DCC, HOBt; (ii) (S)-H-Phe-Me-HCl, E^N; (k) 2,4-bis(phenylthio)-l,3-dithia-2,4-diphosphetane-2,4-disulfide; (1) LiAlH4; (m) (i) DCC, HOBt; (ii) AT-Me-(S)-phenylalanine methyl ester; (n) LiBH.

Scheme 11°



<sup>(</sup>a) Rich, D. H.; Green, J.; Toth, M. V.; Marshall, G. R.; Kent, S. B. H. Hydroxyethylamine Analogues of the pl7/p24 Substrate Cleavage Site are Tight-Binding Inhibitors of HIV Protease. *J. Med. Chem.* 1990, *33,*1285-1288. (b) Yasui, A.; Douglas, A. J.; Walker, B.; Magee, D. F.; Murphy, R. F. Novel C-Terminal Gastrin Antagonists. *Int. J. Pept. Protein Res.*  1990, *35,* 301-305. (c) Rodriguez, M.; Aumelas, A.; Martinez, J. A General route to "Carba" Peptide Bond Replacements: Unequivocal Synthesis of Boc-L-Phe- $\Psi$ (CH<sub>2</sub>-CH<sub>2</sub>)-L-Ala-OH and Boc-L-Phe- $\Psi$ -(CH<sub>2</sub>CH<sub>2</sub>)-D-Ala-OH. *Tetrahedron Lett.* 1990, *31,* 5153-5156. (d) Vara Pasad, J. V. N.; Rich, D. H. Addition of Allylic Metals to Alpha-Aminoaldehydes. Application to the Synthesis of Statine, Ketomethylene and Hydroxyethylene Dipeptide Isosteres. *Tetrahedron Lett.* 1990, *31,*1803-1806. (e) Kaltenbronn, J. S.; Hudspeth, J. P.; Lunof, 1000-1000. (c) Makembronn, 0. S., Huuspelli, 0. I., Lun-<br>new F. A.; Michniewicz, D. M.; Nicolades, F. D.; Benine, J. T.;  $R(S, E, A_i, M)$  Michinewicz, D. M.; Ivicolauss, E. D.; Nephle, J. I.; D.; D.; Es-Roark, W. H.; Stier, M. A.; Tinney, F. J.; Woo, P. K. W.; Essenburg, A. D. Renin Inhibitors Containing Isosteric Replacesenpurg, A. D. Renin Innipitors Containing Isosteric Replace-<br>ments of the Amide Bond Connecting the P<sub>2</sub> and P<sub>2</sub> Sites.<sup>*J.*</sup> *Med. Chem.* 1990, 33, 838-845. (f) Mendre, C; Rodriguez, M.;<br>Med. Chem. 1990, 33, 838-845. (f) Mendre, C; Rodriguez, M.; Med. Chem. 1990, 33, 838-845. (f) Mendre, C.; Rodriguez, M.; Lignon, M-F.; Galas, M-C.; Gueudet, C.; Worms, P.; Martinez, Lignon, M-r.; Galas, M-C.; Gueudet, C.; Worms, P.; Martinez,<br>J. Pharmacological Activity of Cholecystokinin Analogues<br>Medicinin M-t<sup>28</sup> Chr<sup>39</sup> Perius, Eur. J. Ph *186, 213, 200, (g) Iizuka, K.; Kagion, Eur. J. Pharmacol, 1990,*<br>186, 213, 200, (g) Iizuka, K.; Kamii, T.; Hamak, H.; Akahan. 186, 213-222. (g) Iizuka, K.; Kamijo, T.; Harada, H.; Akahane, K.; Kubota, T.; Umeyama, H.; Ishida, T.; Kiso, Y. Orally Potent Renin Inhibitors Derived from Angiotensinogen Transition State: Design, Synthesis and Mode of Action. J. Med. Chem. 1990, 33, 2707-2714. (h) Mendre, C.; Rodriguez, M.; Laur, J.; Aumelias, A.; Martinez, J. Peptide and Pseudopeptide Analogues of Cholecystokinin. Chemical Modifications o

A key consideration in the design of peptide mimetic drugs is the synthesis of selective molecules with a pharmaceutically useful duration of action. This has led to the current study of amide bond replacements.

Many publications describe the synthesis and biological evaluation of peptides possessing amide bond surrogates, but very little has been reported on the systematic quantitative rationalization of the physicochemical conse- (6) (a) Rich, D. H.; Green, J.; Toth, M. V.; Marshall, G. R.; Kent, quences of main chain amide bond replacement in terms of lipophilicity, hydrogen bonding, or steric properties. Furthermore, there are very few examples in the literature where a single amide bond has undergone multiple replacements to give several derivatives that have been evaluated in the same in vitro bioassay. One such example is the  $Nle^{28}$ -Gly<sup>29</sup> region of CCK which has been modified by Martinez.<sup>6h</sup>

> As part of our long-term strategy aimed at analyzing the structure-activity relationships (SAR) of amide bond mimetics in neuropeptides it was decided to replace the central amide in the series of CCK-B "dipeptoids" wth several known amide surrogates. This paper describes the synthesis and CCK-A and CCK-B receptor binding affinities of these compounds.

<sup>(7) (</sup>a) Horwell, D. C; Hughes, J.; Hunter, J. C; Pritchard, M. C; Richardson, R. S.; Roberts, E.; Woodruff, G. N. Rationally Designed "Dipeptoid" Analogues of CCK. Alpha-Methyltryptophan Derivatives as Highly Selective and Orally Active Gastrin and CCK-B Antagonists with Potent Anxiolytic Properties. *J. Med. Chem.* 1991, *34,*404-414. (b) Hughes, J.; Boden, P.; Costall, B.; Domeny, A.; Kelly, E.; Horwell, D. C; Hunter, J. C; Pinnock, R. D.; Woodruff, G. N. Development of a Class of Selective Cholecystokinin Type B Receptor Antagonists having Potent Anxiolytic Activity. *Proc. Natl. Acad. Sci. U.S.A.* 1990, *87,* 6728-6732.





 $^a$ (a) Reference 7; (b) LiBH<sub>4</sub>, Me<sub>3</sub>SiCl; (c) RCOCl, Et<sub>3</sub>N; (d) LiOH; (e) Lawesson's reagent.



<sup>a</sup>(a) (i) DCC; (ii) PhCH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub> (for 30) or PhCH<sub>2</sub>CH<sub>2</sub>SH (for 31) or PhCH<sub>2</sub>CH<sub>2</sub>NHMe (for 32); (b) (i) DCC, DMAP; (ii) 33; (c) (i) DCC; (ii) 35.

### **Chemistry**

Two compounds from the series of tryptophanylphenethylamide CCK-B dipeptoids have been selected for the amide replacement strategy. These are compound 4 (CCK-B  $IC_{50} = 852$  nM), which contains a hydroxymethylene side chain, and compound 23 (CCK-B  $IC_{50}$  =  $32$  nM),<sup>7a</sup> which contains an  $\alpha$ -methyl substituent on the tryptophan moiety. These compounds were selected because their CCK-B receptor affinity is such that quantitative binding data can be determined for higher or lower affinity derivatives and because the chemical synthesis of derivatives of the amide groups was considered to be feasible. Both the  $\alpha$ -methyl and the hydroxymethylene groups have been shown previously to be desirable for  ${\rm \widetilde{CCK}\text{-}B}$  receptor binding affinity.<sup>74,8</sup> The substituent

**Scheme** V



"(a) LiAlH4; (b) TPAP, W-methylmorpholine *N-oxide;* (c) (i) Ph3P, BrCH2CH2CH2Ph; (ii) NaH; (d) 10% Pd-C, H2, 40 psi; (e) Mg,  $BrCH_2CH_2CH_2Ph.$ 

**Scheme VI<sup>0</sup>**



*"* (a) Lawesson's reagent.

selected for the N-terminus was (2-adamantyloxy)carbonyl (2-adoc) which has previously been shown to be optimal.<sup>7a,9</sup> The amide bond in compounds 4 and 23 has been replaced as outlined in Schemes I-VI. The reversed amide analogue,  $\psi(NHCO)$  (8 and 9, Scheme I), is prepared from the  $R$ -Trp derivative 3 via a Curtius rearrangement which gives the racemic bis urethane 7. Hydrogenolysis of 7 and coupling with an active ester of the carboxylic acid 21 (Scheme II) followed by treatment with LiOH produces the desired reversed amide derivative as a mixture of two racemic diastereoisomers which are separated into 8 and 9 each of which is a single racemic diastereoisomer of undetermined stereochemistry. The intermediate acid 21 is prepared from 3-phenylpropanoic acid via a two-step procedure involving hydroxymethylation with formaldehyde to give 20, followed by acetylation of the resulting alcohol (Scheme II).

The method used to prepare the ester replacements,  $\psi$ (COO), 11 (Scheme I), 30, 34, and 36 (Scheme IV) involves coupling (S)-S-phenyllactic acid, 2-phenylethanol, alcohol 33, or alcohol 35, respectively, with the appropriate Trp derivative (3 or 22). In the case of 11 the hydroxy-

methylene side chain is obtained by reduction of the carboxylic acid of compound 10. Compounds 34 and 36 contain a succinic or fumaric acid derived side chain which has previously been shown to confer high CCK-B binding affinity.<sup>7a</sup> The thiolester,  $\psi$ (COS), 31 is prepared by an analogous procedure using phenethyl mercaptan.

The methyleneamino replacement,  $\psi(\rm CH_2NH)$ , 24 (Scheme III) is prepared from compound 23 by direct reduction with  $LiBH<sub>4</sub> - Me<sub>3</sub>SiCl.<sup>10</sup>$  Acylation of 24 with acetyl chloride, ethyl succinyl chloride, and ethyl pimeloyl chloride gives the amides 25, 26, and 27, respectively. Treatment of 26 with LiOH gives the acid 28. In the other series the methyleneamino compound 14 (Scheme I) is obtained by reductive animation of the aldehyde 13 with (S)-2-amino-3-phenyl-l-propanol. The aldehyde 13 is prepared via 12.

The N-methylamide,  $\psi$ (CONMe), 32 (Scheme IV) is prepared via direct coupling of  $N$ -methylphenethylamine to the carboxylic acid 22. The non- $\alpha$ -methyl derivative 19 (Scheme I) is prepared by coupling the acid 3 with *N*methyl-(S)-phenylalanine methyl ester to give 18 followed by reduction with lithium borohydride to give 19. The NMR spectrum of 19 is complicated by the presence of amide rotamers.

The aldehyde 39 (Scheme V) is obtained by reduction of the methyl ester 37 to the alcohol 38 and subsequent oxidation to 39. This compound is a key intermediate for the preparation of several of the required amide replacements. The hydroxyethylenes,  $\psi$ (CHOHCH<sub>2</sub>), 42 and 43 are obtained as optically active diastereoisomers via a Grignard reaction. The cis olefin,  $\psi(HC=CH)$ , 40 is obtained by a Wittig reaction of the aldehyde 39 and 1 bromo-3-phenylpropane. Subsequent hydrogenation gives the ethylene replacement,  $\psi(\text{CH}_2\text{CH}_2)$ , 41.

The thioamide,  $\sqrt{\text{CSNH}}$ , 29 (Scheme III) Is prepared by treatment of the amide 23<sup>7a</sup> with Lawesson's reagent. Under the same conditions, the hydroxymethyl derivative 44 (Scheme VI) (CCK-B IC<sub>50</sub> = 6.4 nM)<sup>7a</sup> gives the 4,5-

<sup>(8)</sup> Horwell, D. C; Birchmore, B.; Boden, P. R.; Higginbottom, M.; Ping Ho, Y.; Hughes, J.; Hunter, J. C; Richardson, R. S. Alpha-Methyl Tryptophanylphenylalanines and their Arylethylamine "Dipeptoid" Analogues of the Tetrapeptide Cholecystokinin (30-33). *Eur. J. Med. Chem.* **1990,** *25,* 53-60.

<sup>(9)</sup> Horwell, D. C; Pritchard, M. P.; Richardson, R. S.; Roberts, E.; Aranda, J. Preparation of N-Substituted Cycloalkyl and Polycycloalkyl alpha-Substituted Tryptophanylphenylalanine Analogs as Drugs. Eur. Patent 405537, 2 January 1991 (Warner-Lambert Company); *Chem. Abstr.* **1991,***115(5),* 50307u.

<sup>(10)</sup> Giannis, A.; Sandhoff, K. LiBH<sub>4</sub>(NaBH<sub>4</sub>)/Me<sub>3</sub>SiCl an Unusually Strong and Versatile Reducing Agent. *Angew. Chem. Int. Ed. Engl.* **1989,** *28,* 218-220.

**Table I.** Physical Data of New Compounds

compd	mol formula	mp, °C	anal.
3	$C_{22}H_{26}N_2O_4$	208-209	C, H, N
4	$C_{31}H_{37}N_3O_4.0.5H_2O$	90–92	C, H, N
5	$C_{31}H_{35}N_3O_2S$	59-63	C, H, N
6	$C_{31}H_{35}N_3O_2S$	56–63	C,H,N,S
7	$C_{29}H_{32}N_4O_6$	148-149	C,H,N
8	$C_{31}H_{37}N_3O_4.0.25H_2O$	185-187	C,H,N
9	$\rm C_{31}H_{37}N_3O_4$ 0.2CCl <sub>4</sub>	$77 - 85$	C,H,N
10	$C_{31}H_{34}N_2O_6$	$77 - 84$	C, H, N
11	$C_{31}H_{36}N_2O_5.0.25H_2O$	$48 - 58$	C, H, N
12	$C_{24}H_{31}N_3O_4$	$72 - 80$	C,H,N
13	$C_{22}H_{26}N_2O_3.0.5H_2O$	$50 - 54$	C, H, N
14	$C_{31}H_{39}N_3O_3.0.2H_2O$	$59 - 61$	C, H, N
15	$C_{32}H_{37}N_3O_5$	$74 - 76$	C,H,N
16	$C_{32}H_{37}N_3O_4S$	$63 - 67$	C,H,N,S
17	$C_{31}H_{37}N_3O_3S 0.65EtOAc$	$71 - 77$	C, H, N
18	$C_{33}H_{39}N_3O_5$	$62 - 67$	C,H,N
19	$\rm{C_{32}H_{39}N_3O_4}$ -0.85CHCl <sub>3</sub>	97–99	C, H, N
20	$C_{10}H_{12}O_3$	$71 - 73$	C,H
21	$C_{12}H_{14}O_4$	a	C, H
24	$C_{31}H_{39}N_3O_2 \cdot C_7H_8SO_3 \cdot 0.75H_2O$	90-93	C, H, N
25	$C_{33}H_{41}N_3O_3.0.25H_2O$	$101 - 103$	C, H, N
26	$C_{37}H_{47}N_3O_5$	$67 - 70$	C, H, N
27	$C_{40}H_{53}N_3O_5$	49-54	C, H, N
28	$C_{35}H_{43}N_3O_5.0.25H_2O$	96-99	C, H, N
29	$C_{31}H_{37}N_3O_2S$	134-136	C, H, N
30	$C_{31}H_{36}N_2O_4$	$57 - 63$	C,H,N
31	$C_{31}H_{36}N_2O_3S$	$62 - 65$	C,H,N
32	$C_{32}H_{39}N_3O_3$	$90 - 95$	C,H,N
33	${\rm C}_{13} {\rm H}_{17}$ NO <sub>4</sub>	$59 - 61$	C, H, N
34	$C_{36}H_{43}N_3O_7.0.5H_2O$	69-71	C, H, N
35	$C_{13}H_{15}NO_4$	75–77	C, H, N
36	$C_{36}H_{41}N_3O_7$	96-99	C, H, N
38	$C_{23}H_{30}N_2O_3.0.5H_2O$	$72 - 74$	C, H, N
39	$C_{23}H_{23}N_2O_3$	178-179	C, H, N
40	$C_{32}$ $H_{38}N_2O_2$	$49 - 52$	C,H,N
41	$C_{32}H_{40}N_2O_2.0.25H_2O$	$43 - 46$	C, H, N
42	$C_{32}H_{40}N_2O_3$	$82 - 85$ $75 - 77$	C, H, N C, H, N
43 45	$C_{32}H_{40}N_2O_3.0.5H_2O$	$73 - 76$	C, H, N
	$C_{32}H_{37}N_3O_2S$		

° Compound obtained as a viscous oil.

dihydro-1,3-thiazole 45.<sup>11</sup> The corresponding compound in the des- $\alpha$ -methyl series is obtained by treating the amide 4 with Lawesson's reagent to give the 4,5-dihydro-l,3 thiazole as a mixture of two optically active diastereoisomers with unknown configuration (5, 6) (Scheme I) which are separated by chromatography. The thioamide 17 is prepared in two steps from the amide 15 using 2,4-bis- (phenylthio)-1,3-dithia-2,4-diphosphetane 2,4-disulfide<sup>12</sup> to give the ester 16 which is reduced to the required alcohol 17 by treatment with lithium aluminum hydride.

Physical data for new compounds 3-21, 24-36, 38-43, and 45 are presented in Table **I.** 

### **Results and Discussion**

The CCK-B and CCK-A in vitro receptor binding data for amide replacements incorporated into compounds 4 and 23 are shown in Tables II and III, respectively. The data obtained with the CCK-B antagonist CI-988 (1)<sup>7a</sup> are given for comparison.

In order to probe the physicochemical consequences of replacing an amide bond some molecular modeling using  $N$ -methylacetamide as a reference compound and the derivatives listed in Table IV has been performed. Three physicochemical parameters of the amide bond and amide replacements were selected for this study, lipophilicity,

steric properties, and hydrogen bonding, since these are known to be involved in the molecular recognition phenomena of drug receptor interactions.<sup>13</sup> The data are presented in Table IV. The four columns to the left summarize the steric data generated with the molecular modeling package SYBYL<sup>14</sup> in terms of the distances and angles between the two methyl groups and the volume of space occupied by the entire molecule. These data show that most of the amide replacements are of similar size, and this justifies the use of the term "isostere". Column five contains log *P* data which monitor the lipophilicity of the molecule in the region of the amide bond. These were calculated using the CLOG P<sup>15</sup> program and indicate how the lipophilicity changes "locally" in the region of the amide bond when this is replaced. It should be noted that in this reference compound, N-methylacetamide, the reversed amide replacement is identical to the parent compound. In a less symmetrical molecule such as a drug or a peptide this would not be the case. The final two columns of Table IV give an indication of the hydrogen bond donating (log  $K_a$ ) and accepting (log  $K_b$ ) abilities. These figures are taken from the work of Morris and Taylor and igures are taken from the work of morths and Taylor and<br>co-workers<sup>16</sup> who quantified hydrogen bonding by measuring the association constants between test compounds and a standard proton donor, 4-nitrophenol, or a standard acceptor, *N*-methylpyrrolidinone. Thus, an amide (log  $K_g$ )  $= 3.0$ ) is a much better H-bond acceptor than an ester (log  $K_{\beta} = 1.5$ .

A qualitative comparison of peptide backbone modifications has been published previously,<sup>2</sup> and some of the physicochemical properties of the amino acid side chains have been parametrized.<sup>17</sup> To our knowledge, this is the first time that these properties of the amide bond have been quantified in a single study incorporating such a variety of amide replacements. The data reported in this paper should contribute to a more rational selection of amide bond replacements in other substrates.

The  $N$ -methylacetamide reference compound is a simpler molecule that most amides in peptides or drugs, and conformational properties have not been considered. However, the data in Table IV do provide a useful aid for considering the SAR of the CCK ligands reported in this study (Tables II and III). For example, the amide bond of compound 4 (CCK-B  $IC_{50} = 852$  nM, CCK-A  $IC_{50} =$ 1080 nM, Table II) will tolerate being replaced with a reversed amide, compounds 8 and 9 (CCK-B  $IC_{50} = 1230$ , 780 and CCK-A  $IC_{50} = 317$ , 976 nM, respectively) or an ester, compound 11 (CCK-B IC<sub>50</sub> = 1890 nM, CCK-A IC<sub>50</sub>  $= 2950$  nM) or an N-methylamide, compound 19 (CCK-B)  $IC_{50} = 1460 \text{ nM}, \text{CCK-A IC}_{50} 1080 \text{ nM}) \text{ (Table II) with only}$ a 3-fold change in the CCK-B or CCK-A binding affinities. These replacements change significantly the hydrogen bonding properties and/or the CLOG p values and/or the molecular volume of the amide, respectively (Table IV). Another replacement which increases the molecular vol-

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**Table II.** CCK-B and CCK-A Receptor Binding Data for Amide Replacements Incorporated into Compound 4





<sup>a</sup>IC<sub>50</sub> is the concentration (nM) producing half-maximal inhibition of specific binding of  $[$ <sup>125</sup>I]CCK-8 to CCK receptors in the mouse cerebral cortex (CCK-B) or the rat pancreas (CCK-A). The values given are the geometric mean and the range from at least three separate experiments. <sup>b</sup> These 4,5-dihydro-1,3-thiazole derivatives do not possess the CH<sub>2</sub>OH group. See Scheme I for full structure. <sup>c</sup>8 and 9 were separated by column chromatography. Each is a racemic diastereoisomer of unknown stereochemistry [i.e. each is either  $(R,R$  and  $S,S$ ) or *(R,S* and *S,R)*]. <sup>*d*</sup>Previously reported compound, ref 7a.

**Table III.** CCK-B and CCK-A Receptor Binding Data for Amide Replacements Incorporated into Compound 23





<sup>a</sup> Binding affinities defined in footnote a, Table II. <sup>b</sup> Previously reported compound, ref 7a. *<sup>c</sup>* Olefin has *Z* configuration. <sup>a</sup> Stereochemistry of the chiral quaternary carbon atom is designated S because priority of the substituents has changed. *'* 42 and 43 are optically active diastereoisomers with different but unknown configuration of the chiral carbon in hydroxyethylene replacement. 'Olefin has *E* configuration. \*See Scheme VI for full structure of this 4,5-dihydro-l,3-thiazole derivative.

ume and has similar CCK-B affinity despite its reduced hydrogen bonding acceptor ability is the thioamide 17  $(\text{CCK-B IC}_{50} = 1590 \text{ nM}, \text{CCK-A IC}_{50} = 1980 \text{ nM}).$ 

The methyleneamino group is one of the most widely used replacements for peptidic amide bonds.<sup>18</sup> It is found in renin inhibitors<sup>18a</sup> and its incorporation into bombe- $\sin^{18b}$  gastrin,<sup>18c</sup> substance  $P$ ,<sup>18d</sup> growth hormone releasing  $factor(GRF)$ ,<sup>18e</sup> and secretin<sup>18f</sup> pseudopeptides has led to the identification of receptor antagonists for these peptides. Within our series of CCK-ligands the methyleneamino replacement 14, which is anticipated to be charged at physiological pH and therefore have very different electronic properties to the amide, has much weaker affinity for the CCK-B receptor than the parent amide 4, but CCK-A affinity is retained, resulting in a CCK-A/ CCK-B selectivity of 9-fold (CCK-B  $IC_{50} = 11500$  nM,

CCK-A  $IC_{50} = 1250$  nM).

The 4,5-dihydro-l,3-thiazole derivatives 5 and 6 have little effect on the CCK-B affinity (CCK-B  $IC_{50} = 2580$ , 2510 nM), but one of the stereoisomers has significantly enhanced CCK-A affinity (CCK-A  $IC_{50} = 125$ , 780 nM, respectively) resulting in a modest  $A/B$  selectivity of 21fold.

The overall conclusion is that in this series of CCK-B ligands the central amide bond can be replaced with groups which significantly change the physicochemical properties shown in Table IV without having much effect on CCK-B binding affinity, indicating that this amide group itself does not participate in a binding interaction with the receptor.

The other compound selected for amide replacement, 23 (CCK-B IC<sub>50</sub> = 32 nM, CCK-A IC<sub>50</sub> = 640 nM, Table

Table IV. Physiochemical Properties of Amide Replacements in N-Methylacetamide (See Text for Details)





<sup>a</sup> Hydrogen bond donating (log  $K_a$ ) and hydrogen bond accepting (log  $K_\beta$ ) parameters are taken from ref 16. <sup>b</sup> Log  $K_\beta$  of isopropylamine. Clefin has trans configuration. <sup>d</sup>Olefin has cis configuration. Value estimated from basicity pK<sub>b</sub>. 'N-Methylation claimed to leave acceptor strength unchanged in ref 16.

Ill), has higher CCK-B receptor affinity than 4. Once again, amide replacement with the thioamide 29 (CCK-B  $IC_{50} = 808 \text{ nM}, \text{CCK-A IC}_{50} = 1500 \text{ nM}$  or the methyleneamino 24 (CCK-B IC<sub>50</sub> = 1080 nM, CCK-A IC<sub>50</sub> = 4060 nM) results in significantly reduced activity compared to the parent amide 23. The amino group of 24 was acetylated in order to restore the amide character to the nitrogen atom. However, the acetate 25 (CCK-B  $IC_{50} = 1060$ nM, CCK-A  $IC_{50} = 1310$  nM) does not restore CCK-B affinity.

Incorporation of a succinic ester or a succinic acid in the CI-988 series has been shown previously to enhance CCK-B affinity.<sup>7a</sup> The same groups were attached to the aminomethylene replacement to give 26 and 28, but they do not increase affinity (CCK-B *lCm* = 1630,1270 nM and CCK-A  $IC_{50} = 670$ , 1080 nM, respectively). The ethyl pimelate side chain was appended to give 27 (CCK-B IC $_{50}$  $=$  863, CCK-A IC<sub>50</sub> = 3530) which contains the same number of bonds between the amide nitrogen and the ester as is found in the CI-988 series. Once again, this fails to restore CCK-B affinity.

The amide in 23 is more sensitive to replacement than that of 4 because, unlike the compounds in Table II discussed above, the  $N$ -methylamide 32, the 4,5-dihydro1,3-thiazole 45, and the ester 30 all have significantly weaker CCK-B receptor affinity (CCK-B  $IC_{50} = 123, 801$ , 547 nM and CCK-A  $IC_{50} = 757, 827, 1070$  nM, respectively). The thiolester 31 (CCK-B IC<sub>50</sub> = 932 nM,  $\rm \AA/B$ ratio  $= 1.3$ ) has similar binding affinity to the ester 30 despite the differences between them in terms of molecular volume  $(76.4 \text{ and } 65.8 \text{ Å}^3 \text{ respectively, Table IV})$ , size, and CLOG p values.

The methyl succinate and methyl fumarate side chains found in the CI-988 series<sup>7a</sup> were appended to the ester 30 to give 34 and 36 which have 8- and 14-fold improved affinity for the CCK-B receptor, respectively. This ester replacement is of particular interest here because it should be stabilized toward esterase hydrolysis by steric hindrance of the  $\alpha$ -methyl group, and it cannot undergo the hydantoin formation which has been shown to occur when the corresponding amides in the CI-988 series are treated with base.<sup>19</sup>

The cis double bond in 40 has different steric, CLOG p, and hydrogen-bonding properties compared to the amide, and it is much less active (CCK-B *IC^* = 6340 nM, CCK-A  $IC_{50} = 3620$  nM). It is the synthetic precursor for the dimethylene replacement 41 which is a closer mimic of the steric properties of the amide in terms of the parameters in Table IV. However, this does not improve affinity (CCK-B IC<sub>50</sub> = 8690 nM, CCK-A IC<sub>50</sub> = 19500 nM), indicating that these steric properties are not the only reason for the inactivity of 40.

The hydroxyethyl group has been used successfully to replace the amide bond in peptide substrates of proteolytic enzymes<sup>4a,6a,6e</sup> where it is thought to mimic the tetrahedral sp<sup>3</sup> transition state of the amide bond hydrolysis that is catalyzed by these enzymes. In this series of CCK-B ligands the hydroxyethyl replacements 42 and 43 are significantly less active than the parent amide (by 19-fold and 62-fold, respectively), and the data in Table IV show that they are poor mimics of the amide bond in terms of lipophilicity, molecular volume, and the bond angles  $\alpha$  and  $\beta$ .

In summary, the amide replacements incorporated into 23 reduce CCK-B binding affinity and CCK-A/CCK-B selectivity. The amide replacements which have been used

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<sup>(19)</sup> Horwell, D. C; Ratcliffe, G.; Roberts, E. Stabilities of Tryptophanylphenethylamides to Acid and Alkaline Conditions. *BioMed. Chem. Lett.* **1991,***1,* 169-172.

as transition-state mimics to increase activity in peptidic enzyme inhibitors (methyleneamino, hydroxyethylene) are particularly inactive (compounds 14,24,42, and 43) presumably because they are poor mimics of the amide ground state which is expected to be the state recognized by the reversible interactions with receptors.

Other replacements which are poor mimics of the amide in terms of steric, hydrogen bonding, and lipophilic properties (cis double bond, ethylene) are also amongst the least active compounds (40, 41). However, some change to the molecular volume of the amide group appears to be tolerated by the CCK-B receptor (e.g. N-methylamide compounds 19, 32). The CCK-A and CCK-B receptors have different binding preferences for this central amide bond and the 4,5-dihydro-l,3-thiazole replacement 5 has a significantly higher affinity and selectivity for the CCK-A receptor than does the parent amide 4 ( $IC_{50} = 125$  and 1080 nM, respectively).

Although many of these replacements reduce the CCK-B receptor affinity, it has been possible to restore this by appending an auxiliary side chain to the phenethyl group. Thus the ester replacement 30 has a 17-fold weaker CCK-B affinity than the amide 23 ( $IC_{50} = 547$  and 32 nM, respectively) but attaching a fumarate side chain gives compound 36 which is approximately equiactive as  $23 \overline{{\rm (IC}_{50}}$  $= 38.8 \text{ nM}$ .

#### **Conclusion**

The central amide bond of the CCK-B ligands 4 and 23 has been replaced with eleven different amide replacements. This represents one of the most comprehensive studies of amide replacements in a ligand for a membrane bound receptor. It has resulted in a series of compounds which generally have weaker CCK-B receptor affinity than the parent amide. This decrease in affinity indicates that these replacements are poor mimics of the amide bond in terms of CCK-B receptor binding interactions with these ligands. Since no correlation between the binding affinity and the physicochemical properties of the compounds in Table IV has been determined, it is presumed that other factors (e.g. conformation) are of greater importance for this particular ligand-receptor interaction. These results are contrasted with many successful reports in the literature of the replacement of amide bonds in certain enzyme substrates to produce potent enzyme inhibitors (e.g. renin, angiotensin converting enzyme, HIV proteinase). It should be noted that these enzymes are thought to interact with a transition state of amide bond hydrolysis but the receptors presumably bind to a ground state of the amide.

#### **Experimental Section**

**Molecular Modeling.** The models of N-methylacetamide, and its amide-replaced analogues, were built using the commercial package SYBYL (version 5.3), running on a Silicon Graphics 4D-310 workstation. Each model was built to mimic the trans configuration of the amide as closely as possible. The geometry was then optimized using MAXIMIN II (A)<sup>20</sup> with the Tripos force field.<sup>21,22</sup> No atomic charges were calculated. The molecular volumes were calculated as that space within one van der Waals radii of the

atoms, using the surfaces/volume option, CLOG P values were calculated according to the method of Leo.<sup>15</sup>

Melting points were determined with a Reichart Thermovar hot-stage apparatus and are uncorrected. All NMR spectra were *<sup>l</sup>H* NMR recorded on a Bruker AM300 MHz spectrometer or a JEOL PMX60 MHz spectrometer; chemical shifts were recorded in parts per million downfield from tetramethylsilane and coupling constants *(J)* are measured in hertz. IR spectra were recorded with compound (neat) on a sodium chloride disk with a Perkin-Elmer 1750 Fourier transform spectrometer. Optical rotations were determined on a Perkin-Elmer 241 polarimeter. Silica gel used for chromatography was Kieselgel-60 (230-400 mesh); reverse-phase silica gel was Lichroprep RP-18 (230-400 mesh). Both were supplied by E. Merck AG, Darmstadt, Germany. Elemental analyses were determined by CHN Analyses Limited, Leicester, U.K. Mass spectra were recorded with a Finnegan 4500 spectrometer. All experiments using moisture sensitive reagents were performed under a dry nitrogen atmosphere.

**Tricyclo[3.3.1.1"]dec-2-yl[l-Carboxy-2-(l#-indol-3-yl)** ethyl]carbamate (3). Tricyclo<sup>[3.3.1.13,7</sup>]dec-2-yl chloroformate<sup>23</sup> (2.97 g, 13.9 mmol) in 1,4-dioxane (10.5 mL) was added dropwise with stirring to a cooled (ice bath) mixture of  $(R)$ -tryptophan  $(2.13)$ g, 10.4 mmol),  $NAHCO<sub>3</sub>$  (879 mg, 10.5 mmol), and 1 N NaOH (10.5 mL, 10.5 mmol) in 1,4-dioxane (10.5 mL). After the addition was complete, the cooling bath was removed, and the mixture was stirred at room temperature for 18 h. The 1,4-dioxane was removed in vacuo, and the residue was partitioned between 10% citric acid (100 mL) and ethyl acetate (100 mL). The layers were separated, and the aqueous phase was extracted with ethyl acetate  $(3 \times 50 \text{ mL})$ . The combined organic phase was washed with brine (lOO mL), dried (MgS04), and concentrated in vacuo. The crude product was crystallized from ethyl acetate to give the acid 3 (2.03 g, 51%):  $[\alpha]_{D}^{22}$  +9.4° (c = 1.2, MeOH); IR (film) 3399 (OH), 2908 and  $2854$  (CH),  $1715$  sh (CO acid), and  $1693$  (CO urethane) cm<sup>-1</sup>; NMR (DMSO- $d_6$ )  $\delta$  1.31-2.09 (14 H, m, adamantyl), 3.07 (1 H, dd, *J* = 14,9 Hz), 3.18 (1H, dd, *J* = 14,5 Hz), 4.22 (1 H, m), 4.58 (1 H, s), 6.99 (1 H, dd, *J* - 8, 8 Hz), 7.12 (1 H, dd, *J* = 8, 8 Hz), 7.17 (1H, s), 7.25 (1H, d, *J* = 9 Hz), 7.34 (1 H, d, *J* = 8 Hz, indole 7-H), 7.54 (1 H, d, *J* = 8 Hz, indole 4-H), 10.84 (1 H, s), 12.62  $(1 \text{ H, br s})$ ; MS  $m/e$  (CI) 383 (M<sup>+</sup> + H, 31%).

**Tricyclo[3.3.1.1<sup>8</sup> - 7 ]dec-2-yl[fi-(ii\*^\*)]-[2-[[l-(Hydroxymethyl)-2-phenylethyl]amino]-l-(lff-indol-3-ylmethyl)-2 oxoethyl]carbamate (4).** The carboxylic acid 3 (380 mg, 1.0 mmol), N,N'-dicyclohexylcarbodiimide (DCC) (230 mg, 1.1 mmol), and pentafluorophenol (PFP) (200 mg, 1.1 mmol) were suspended in EtOAc (25 mL), stirred for 2 h, and then treated with (S)-2 amino-3-phenyl-l-propanol (150 mg, 1.0 mmol) and stirred at 40 °C for 18 h. The mixture was filtered, and the filtrate was washed with saturated aqueous citric acid, followed by saturated aqueous NaHCO<sub>3</sub> and then H<sub>2</sub>O. The organic layer was dried (MgSO<sub>4</sub>), filtered, concentrated in vacuo, and purified by reverse-phase silica gel chromatography, using MeOH-H20 (4:1) as eluant to give the product 4 (0.36 g, 71%):  $[\alpha]^{\infty}$ <sub>D</sub> = 20.4° (c = 0.25, CH<sub>2</sub>Cl<sub>2</sub>); NMR  $(CDCI<sub>3</sub>)$   $\delta$  1.5-2.0 (14 H, m), 2.20 (1 H, br s), 2.60 (2 H, d, J = 7 Hz), 3.1-3.5 (4 H, m), 4.0-4.1 (1 H, m), 4.42 (1 H, q, *J* = 7 Hz), 4.6-4.8 (1 H, m), 5.45 (1 H, br s), 6.15 (1 H, br s), 6.9-7.3 (8 **H,**  m), 7.37 (1 **H,** d, *J* = 8 Hz), 7.67 (1 H, d, *J* = 8 Hz), 8.25 (1 H, brs).

**Tricyclo[3.3.1.1<sup>87</sup>]dec-2-yl [l-[4,5-Dihydro-4-(phenylmethyl)-2-thiazolyl]-2-(lH-indol-3-yl)ethyl]carbamate** (5 **and 6).** The hydroxyamide 4 (339 mg, 0.657 mmol) and Lawesson's reagent (Aldrich) (250 mg, 0.618 mmol) were refluxed in toluene (10 mL) for 1 h. The solution was cooled to room temperature and purified by column chromatography using ether-hexane (60:40) to give the diastereoisomeric thiazolines 5 and 6. Isomer I (57 mg, 17%):  $[\alpha]^{20}$ <sub>D</sub> = 9.1° (c = 0.52, MeOH); IR (film) 2911, 2854, 1702, and 1620 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  1.44-2.11 (14 H, m), 2.42 (1 H, m), 2.96 (2 H, dd, *J* = 7 and 11 Hz), 3.14 (1 H, dd, *J*   $= 8$  and 11 Hz), 3.21-3.47 (2 H, m), 4.65 (1 H, m), 4.84 (1 H, s), 4.89 (1 H, br s), 5.57 (1 H, br d,  $J = 7$  Hz), 6.92-7.35 (9 H, m), 7.68 (1 H, d, *J* = 8 Hz), 8.12 ( 1 H, s); MS *m/e* (CI) 514 (M<sup>+</sup> + H) (100%). Isomer II (170 mg, 50%):  $[\alpha]_{D}^{\infty} = -31^{\circ}$  (c = 1.00,

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<sup>(22)</sup> Vinter, J. G.; Davis, A.; Saunders, M. R. Strategic Approaches to Drug Design. 1. An Integral Software Framework for Molecular Modelling. *J. Comput.-Aided Mol. Des.* **1987,***1,* 31-51.

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CHCl<sub>3</sub>); IR (film) 2909, 2854, 1698, and 1621 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) 8 1.43-2.09 (14 H, m), 2.68 (1 H, dd, *J* = 9 and 14 Hz), 2.83-3.44 (5 H, m), 4.62 (1 H, m), 4.82 (1 H, s), 4.88 (1 H, br s), 5.44 (1 H, br d, *J* = 7 Hz), 6.96-7.38 (9 **H,** m), 7.63 (1 **H, d,** *J* = 8 Hz), 8.16 (1 **H,** s); MS *m/e* (FAB) 514 (M<sup>+</sup> + **H)** (19%).

**Tricyclo[3.3.1.1<sup>3</sup> ' 7 ]dec-2-yl(±)-[2-(lff-Indol-3-yl)-l-[[[(4 nitrophenyl)methoxy]carbonyl]amino]ethyl]carbamate** (7). To a solution of acid 3 (3.62 g, 9.47 mmol) in THF (36 mL) at  $-10$  °C was added N-methylmorpholine (1.15 mL, 10.4 mmol) and isobutyl chloroformate (1.35 mL, 10.4 mmol). This mixture was stirred for 20 min at -10 °C and then filtered. Trimethylsilyl azide (Aldrich) (1.89 mL, 14.2 mmol) was added to the filtrate, and the resulting solution was stirred at -10 °C for 1 h. The solvent was removed in vacuo at 25 °C, and the residue was partitioned between EtOAc (100 mL) and saturated aqueous  $NaHCO<sub>3</sub>$  (100 mL). The organic phase was washed with brine, dried  $(MgSO<sub>4</sub>)$ , and concentrated in vacuo at 25 °C. The residue was dissolved in toluene (100 mL) and heated at 40 °C until rearrangement to the isocyanate was complete  $(30 \text{ min})$ . [IR (neat)  $2139 \text{ (N}_3)$ ;  $2249$ (NCO) cm"<sup>1</sup> ]. p-Nitrobenzyl alcohol (2.20 g, 14.3 mmol) and l,4-diazabicyclo[2.2.2]octane (DABCO) (149 mg, 1.33 mmol) were added, and the mixture was left at 40 °C for 15 h. The solvent was removed in vacuo, and the crude product was purified by silica gel chromatography using ether-hexane (4:1) as eluant to give 7 as a yellow solid which was recrystallized from ether-hexane (2.12 g, 42%): IR (film) 2908, 2855 (adamantyl C-H), 1703 (br, CO urethane), 1520 (NO<sub>2</sub>), and 1347 (NO<sub>2</sub>) cm<sup>-1</sup>; NMR (DMSO- $d_6$ )  $\delta$  1.39-2.03 (14 H, m), 3.07 (2 H, br s, CH<sub>2</sub>-indole), 4.64 (1 H, s, adamantyl H-2), 5.14 (2 H, s,  $CH_2Ph$ ), 5.37 (1 H, m, CH- $(NHCO<sub>2</sub>)NHCO<sub>2</sub>$ ), 6.90-7.29 (4 H, m, indole H-2, H-5, H-6 NH), 7.34 (2 H, d, *J* = 8 Hz, PhH-2, PhH-6), 7.46-7.62 (3 H, m, indole H-4, H-7, NH), 8.16 (2 H, d, *J* = 8 Hz, PhH-3, PhH-5), 10.61 (1 H, s, indole NH).

 $\text{Tricyclo[}3.3.1.1^{3.7}]\text{dec-}2\text{-yl }(\pm)$ -[1-[[2-(Hydroxymethyl)-1 **oxo-3-phenylpropyl]amino]-2-(lff-indol-3-yl)ethyl]carbamate** (8 **and 9).** The bis urethane 7 (524 mg, 0.984 mmol) was dissolved in EtOAc (36 mL) and treated with 20% palladium hydroxide on carbon (Pearlman's catalyst). The mixture was put under an atmosphere of hydrogen gas at 45 psi at 40 °C for 1 h and then filtered through Celite (to remove the catalyst) into a flask containing an activated ester of 21 which had been generated via the reaction of 21 (226 mg, 1.02 mmol) in EtOAc (5 mL) with 1-hydroxybenzotriazole hydrate (HOBt) (179 mg, 1.17 mmol) and  $N$ . V'-dicyclohexylcarbodiimide (235 mg, 1.14 mmol) at 0 °C for 90 min. The resulting mixture was stirred at room temperature for 18 h, then concentrated in vacuo, cooled to 0 °C, filtered, and evaporated to dryness. The crude product was purified by silica gel chromatography using CH<sub>2</sub>Cl<sub>2</sub>-EtOAc (80:20) as eluant to give the acetate (265 mg, 48%): IR (film) 2916,2855 (adamantyl), 1740 sh (CO ester),  $1698$  (CO urethane), and  $1658$  (CO amide) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.47-2.14 (17 H, m, adamantyl, CH<sub>3</sub>), 2.62 (1 H, m, *COCH),* 2.78 (1 H, m, CH of CH2Ph), 2.90 (1 H, m, CH of  $CH<sub>2</sub>Ph$ ), 3.13 (1 H, m, CH of  $CH<sub>2</sub>$ -indole), 3.23 (1 H, m, CH of  $CH<sub>2</sub>$ -indole), 4.18 (2 H, m,  $CH<sub>2</sub>OCO$ ), 4.80 (1 H, s, adamantyl H-2), 5.41 (1 H, br d, *CHNH),* 6.25 (0.5 H, br s), 6.77 (0.5 H, s, indole H-2), 6.88 (0.5 H, s, indole H-2), 7.04-7.62 (9.5 H, m, Ph, indole H-4, H-5, H-6, H-7), 8.09 (1 H, s, indole NH); MS *m/e* (FAB) 558 (3)  $(M^+ + 1)$ , 363 (83)  $(M^+ - C_{11}H_{17}NO_2)$ , 135 (100)  $(C_{10}H_{16})$ . This acetate (264 mg, 0.473 mmol) was dissolved in THF-MeOH-H<sub>2</sub>O (3:2:1) (6 mL) and treated with lithium hydroxide monohydrate (56.6 mg, 1.35 mmol) at room temperature for 30 min. The reaction mixture was poured into 2 N HC1 (50 mL) and extracted with EtOAc  $(3 \times 25 \text{ mL})$ . The EtOAc fractions were combined and washed with brine, dried  $(MgSO<sub>4</sub>)$ , and evaporated to dryness. The crude material was purified by silica gel chromatography using  $CH<sub>2</sub>Cl<sub>2</sub>$ -MeOH (95:5) as eluant to give the diastereoisomeric alcohols (8 and 9) (150 mg, 62%). Diastereoisomer I, 8, (80 mg): IR (film) 3540-3140 (br, OH), 2910, 2855 (adamantyl C-H), 1695 (CO urethane), and 1660 (CO amide) cm<sup>-1</sup>; NMR (CD<sub>3</sub>OD)  $\delta$ 1.43-2.04 (14 H, br m, adamantyl), 2.42 (1 H, m, NHCOCH), 2.67-3.22 (5 H, br m, indole CH<sub>2</sub>, PhCH<sub>2</sub>, OH), 3.67 (2 H, m, CH<sub>2</sub>OH), 4.78 (1 H, s, adamantyl H-2), 5.40 (1 H, br s), 5.58 (1 H, br s), 6.26 (1 H, br s, amide NH), 6.82 (1 H, s, indole H-2), 7.02-7.37 (8 H, m, Ph, indole H-5, H-6, H-7), 7.44 (1 H, d, *J* = 8 Hz, indole H-4), 8.16 (1 H, s, indole NH); MS *m/e* (FAB) 538 (39), 516 (8) (M<sup>+</sup> + H), 321 (100). Diastereoisomer II, 9 (70 mg): IR (film) 3440 (br, OH), 2908, 2856 (adamantyl C-H), 1698 (CO

urethane), and 1657 (CO amide) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  1.47-2.14 (14 H, m, adamantyl), 2.44 (1 H, m, *COCH),* 2.74 (1 H, dd, *J =*  7, 14 Hz, one of  $CH_2Ph$ , 2.84 (1 H, dd,  $J = 8$ , 14 Hz, one of *CH*<sub>2</sub>Ph), 2.95 (1 H, br s, OH), 3.19 (2 H, d,  $J = 5$  Hz, CH<sub>2</sub>-indole), 3.64 (2 H, s, CH<sub>2</sub>OH), 4.79 (1 H, s, adamantyl H-2), 5.27 (1 H, br d, indole-CH<sub>2</sub>CH), 5.49 (1 H, br s, urethane NH), 6.42 (1 H, br s, amide NH), 6.91 (1H, s, indole H-2), 7.02-7.29 (7 H, m, indole H-6, H-5, Ph), 7.35 (1 H, d, *J* = 8 Hz, indole H-7), 7.56 (1 H, d, *J = 8* Hz), 8.30 (1 H, s, indole NH); MS *m/e* (CI) 516 (0.2) (M<sup>+</sup> **+ H),** 321 (94), 180 (100).

 $(S)$ **-N**-[(Tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yloxy)carbonyl]- $(R)$  **tryptophan, l-Carboxy-2-phenylethyl Ester (10).** The acid 3 (428 mg, 1.12 mmol) was treated with pentafluorophenol (247 mg, 1.34 mmol),  $N$ , $N'$ -dicyclohexylcarbodiimide (254 mg, 1.23 mmol), 4-(dimethylamino)pyridine (DMAP) (14 mg, 0.11 mmol), and  $(-)$ -(S)-3-phenyllactic acid (208 mg, 1.25 mmol) in DMF (3) mL) and stirred at room temperature for 18 h. Water (21 mL) was added, and the resulting emulsion was extracted with ether  $(50 \text{ mL})$ . The organic phase was washed with  $H<sub>2</sub>O$  and brine, dried (MgS04), filtered, and concentrated in vacuo. The crude material was purified by silica gel chromatography using EtOAc-n-hexane-acetic acid (66:33:1) as eluant to give acid/ester 10 as a white foam (396 mg, 66%): IR (film) 3410 (br, OH), 2913, 2854 (adamantyl C-H), 1736 (br, CO ester/acid), and 1703 (CO urethane) cm<sup>-1</sup>; NMR (CD<sub>3</sub>OD) δ 1.52-2.17 (14 H, m, adamantyl), 2.95-3.30  $(4 \text{ H}, \text{ m}, 2 \times \text{CH}_2)$ , 4.63 (1 H, m, ind CH<sub>2</sub>CH), 4.71 (1 H, s, adamantyl H-2),  $5.18$  (1 H, m, OCHCO<sub>2</sub>H), 6.77 (1 H, s, indole H-2), 6.92-7.39 (8 H, m, indole H-7, H-6, H-5, Ph), 7.48 (1 H, d,  $J = 8$  Hz, indole H-4); MS (FAB)  $m/e$  531 (98) (M<sup>+</sup> + H), 470 (100), 396 (5), 383 (11).

 $(S)$ - $N$ -[(Tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yloxy)carbonyl]- $(R)$ **tryptophan, l-(Hydroxymethyl)-2-phenylethyl Ester** (11). The acid 10 (208 mg, 0.392 mmol), ethyl chloroformate  $(41 \mu L,$ 0.47 mmol), and triethylamine  $(66 \mu L, 0.47 \text{ mmol})$  in THF  $(1 \text{ mL})$ were stirred at 5 °C for 30 min. The mixture was filtered, and the filtrate was added to a cooled solution (ice bath) of sodium borohydride (46 mg, 1.2 mmol) in  $H<sub>2</sub>O$  (2 mL). The solution was warmed to room temperature and stirred for 18 h, then acidified with concentrated HCl to pH 2, and extracted with EtOAc  $(3 \times$ 25 mL). The organic phase was washed with brine, dried  $(MgSO<sub>4</sub>)$ , filtered, and concentrated in vacuo. The crude material was purified by silica gel chromatography using  $CH_2Cl_2-EtOAc$  (1:1) as eluant to give hydroxy ester **11** as a white foam (144 mg, 71%): IR (film) 3418 (OH), 2902,2856 (adamantyl C-H), 1740 (CO ester), and 1697 (CO urethane) cm<sup>-1</sup>;  $[\alpha]^{20}$ <sub>D</sub> = +7.6° (c = 1.01, CHCl<sub>3</sub>); NMR (CDCl<sub>3</sub>) δ 1.44-2.18 (14 H, m, adamantyl), 2.63 (2 H, br s, PhCH2), 3.39 (2 H, br s, indole CH2), 3.77-4.02 (2 H, m, *CH*  of CH2OH, OCffCH2Ph), 4.11 (1H, dd, *J* = 3 and 11 Hz, *CH* of CH2OH), 4.65-4.75 (1 H, m), 4.82 (1 H, s, adamantyl H-2), 5.28 (1 H, br d, *J =* 8 Hz, urethane NH), 6.93-7.38 (9 H, m, indole H-7, H-6, H-5, H-2, Ph), 7.57 (1 H, d, *J* = 8 Hz, indole H-4), 8.16 (1 H, s, indole NH); MS (FAB) *m/e* 516 (100) (M<sup>+</sup> ).

 $Tricyclo[3.3.1.1<sup>3,7</sup>]$ dec-2-yl  $(R)$ -[1-(1*H*-Indol-3-yl**methyl)-2-(methoxymethylamino)-2-oxoethyl]carbamate (12).** To a solution of acid 3 (859 mg, 2.25 mmol) in  $CH_2Cl_2$  (8) mL) was added N-methylmorpholine (495  $\mu$ L, 4.50 mmol). The mixture was cooled to -15 °C, and isobutyl chloroformate (292  $\mu$ L, 2.25 mmol) was added. The mixture was stirred at -15 °C for 15 min and then treated with  $N,O$ -dimethylhydroxylamine hydrochloride (219 mg, 2.25 mmol). The reaction mixture was stirred at -15 °C for 1 h, then warmed to room temperature, and stirred for a further 15 h. The mixture was filtered, and the filtrate was washed sequentially with aqueous  $NAHCO<sub>3</sub>$ , water, 10% citric acid, and brine, dried (MgS04), and evaporated to dryness. The crude material was purified by silica gel chromatography using hexane-EtOAc (1:1) as eluant to give the hydroxamate **12** as a white foam (694 mg, 73%):  $[\alpha]^{\mathcal{W}}_{\mathbf{D}} = -9.4^{\circ}$  (c = 0.30, CHCl<sub>3</sub>); IR (film) 2908, 2855, 1695, and 1651 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  1.30-2.01 (14 H, m, adamantyl), 2.99-3.24 (5 H, m, indole CH<sub>2</sub>, NCH<sub>3</sub>), 3.58  $(3 H, s, OCH<sub>3</sub>)$ , 4.7  $(1 H, s, \text{adamantyl H-2})$ , 5.00  $(1 H, \text{ br } d,$  $CH(NHR)CON$ ), 5.36 (1 H, br d, urethane NH), 6.92-7.17 (3 H, m, indole H-2, H-5, H-6), 7.27 (1H, d, *J* = 8 Hz, indole H-7), 7.53  $(1 H, d, J = 8 Hz,$  indole H-4), 7.99  $(1 H, s,$  indole NH).

**Tricyclo[3.3.1.1"]dec-2-yl(ii)-[l-Formyl-2-(lfr-indol-3 yl)ethyl]carbamate (13).** Lithium aluminum hydride (45 mg, 1.2 mmol) was added portion wise over a period of 30 min to a solution of the hydroxamate **12** (197 mg, 0.460 mmol) in THF (3

mL) at 0 °C. The mixture was stirred for a further 30 min, and then ether (30 mL) was added, followed by an ice-cold solution of 10% citric acid (40 mL). The mixture was stirred vigorously for 30 min, then the layers were separated, and the aqueous phase was extracted with ether  $(5 \times 10 \text{ mL})$ . The ether extracts were combined and sequentially washed with saturated aqueous  $\text{NaHCO}_3$  (25 mL), H<sub>2</sub>O (25 mL), 10% citric acid (25 mL), and brine (25 mL), dried ( $\text{Na}_2\text{SO}_4$ ), and concentrated in vacuo to give the aldehyde 13 as a white foam (140 mg, 83%):  $[\alpha]_{D}^{\infty} = 36.9$  $(c = 1.14, CHCl<sub>3</sub>)$ ; IR (film) 1725 sh (aldehyde CO) and 1693 (urethane CO) cm"<sup>1</sup> ; NMR (CDC13) *6* 1.47-2.13 (14 H, m, adamantyl), 3.26 (2 H, dd,  $J = 7$ , 15 Hz, indole-CH<sub>2</sub>), 4.58 (1 H, br d, Cff(NHR)CHO), 4.84 (1 H, s, adamantyl H-2), 5.3 (1 H), 6.98-7.26 (3 H, m), 7.35 (1 H, d, *J* = 8 Hz, indole H-7), 7.60 (1 H, br d, *J* = 8 Hz, indole H-4), 8.24 (1 H, br s, indole NH), 9.64 (1 H, s, CHO).

Tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yl [ $R-(R^*,S^*)$ ]-[2-[[1-(Hydroxy $methyl-2-phenylethyl] amino-1-(1H-indol-3-ylmethyl-1)$ ethyljcarbamate (14). Sodium cyanoborohydride (37 mg, 0.59 mmol) was added portionwise over a period of 15 min to a solution of the aldehyde 13 (136 mg, 0.370 mmol) and (S)-2-amino-3 phenyl-1-propanol (61 mg, 0.40 mmol) in MeOH-acetic acid (99:1) (5 mL). The mixture was stirred for 2 h at room temperature and then chilled (ice bath). Saturated aqueous  $\mathrm{NaHCO}_{3}\left(30\;\mathrm{mL}\right)$ was added with stirring, followed by EtOAc (45 mL). The organic layer was separated, washed with brine (5 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to dryness. The crude product was purified by silica gel chromatography using  $CH_2Cl_2-MeOH$  (95:5) as eluant to give the amino alcohol 14 as a beige foam (60 mg, 32%):  $[\alpha]^{20}$ <sub>D</sub>  $= +1.2$ ° (c = 1.1, CHCl<sub>3</sub>); IR (film) 3325 (OH), 1690 (CO urethane), 1496 (N-H), 1266 (OH), 1048 (C-O), 740 and 701 (monosubstituted Ph) cm"<sup>1</sup> ; NMR (CDC13) *&* 1.31-2.04 (16 H, m, adamantyl, NH, OH), 2.44-3.01 (7 H, m, indole-CH<sub>2</sub>, CH<sub>2</sub>NH, CH<sub>2</sub>Ph, NHCH(CH<sub>2</sub>OH)CH<sub>2</sub>Ph), 3.21 (1 H, dd,  $J = 6$  and 11 Hz, one of CH<sub>2</sub>OH), 3.47 (1 H, dd,  $J = 4$  and 11 Hz, one of CH<sub>2</sub>OH), 3.97  $(1 \text{ H}, \text{m}, \text{CH}_2\text{CH}(\text{NHCO}_2\text{R})\text{CH}_2\text{NH})$ , 4.66  $(1 \text{ H}, \text{br d}, \text{urethane})$ NH), 4.73 (1 H, s, adamantyl H-2), 6.87 (1 H, d, *J* = 2 Hz, indole H-2), 6.96-7.24 (7 H, m, indole H-5, H-6, Ph), 7.29 (1 H, d, *J =*  8 Hz, indole H-7), 7.55 (1 H, d, *J* = 7 Hz, indole H-4), 7.97 (1 H, br s, indole NH).

 $N$ -[ $N$ -[(Tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yloxy)carbonyl]- $(R)$  tryptophyl]-(S)-phenylalanine, Methyl Ester (15). (S)- Phenylalanine methyl ester hydrochloride (Aldrich) (361 mg, 1.67 mmol) and triethylamine (233  $\mu$ L, 1.67 mmol) were stirred at room temperature in ethyl acetate (10 mL) for 10 min and then filtered. The filtrate was added dropwise with stirring to a cooled (ice/ salt) solution of the active ester of acid 3 [generated by stirring acid 3 (426 mg, 1.11 mmol), *NJ\*J* -dicyclohexylcarbodiimide (228 mg, 1.11 mmol), and 1-hydroxybenzotriazole monohydrate (HOBt) (182 mg, 1.35 mmol) for 10 min in dichloromethane (10 mL) in an ice/salt bath]. The resulting mixture was warmed to room temperature and stirred for 2 h [at which point TLC  $(SiO<sub>2</sub>;$ toluene-acetic acid 90:10) showed that no starting material remained]. The solvent was removed in vacuo, and the residue was stirred in ethyl acetate (20 mL), chilled (ice/salt), and then filtered. The filtrate was washed with saturated sodium hydrogen carbonate, water, 2 N hydrochloric acid, and brine, dried  $(\overline{MgSO_4})$ , and filtered, and the filtrate was concentrated in vacuo. The crude material was purified by silica gel column chromatography using dichloromethane-ethyl acetate (80:20) as eluant and then recrystallized from ether-hexane to give 15 as a white solid (459 mg, 76%):  $[\alpha]_{\text{D}}^{\infty}$  = +3.0° (c = 1.1, MeOH); IR (film) 2907, 2854, 1739, 1700, and 1665 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  1.43-2.13 (14 H, m), 2.82 (1 H, dd, *J* = 6 and 14 Hz), 2.93 (1 H, dd, *J* = 6 and 14 Hz), 3.08-3.42 (2 H, m), 3.61 (3 H, s), 4.53 (1 H, dd, *J* = 7 and 14 Hz), 4.67-4.83 (2 H, m), 5.32 (1 H, br s), 6.19 (1 H, br d, *J* = 6 Hz), 6.74-6.99 (2 H, m), 7.08-7.23 (6 H, m), 7.34 (1 H, d, *J* = 8 Hz), 7.66 (1 H, d, *J* = 7 Hz), 8.08 (1 H, s); MS *m/e* (CI) 544 (100) (M<sup>+</sup> + H), 409 (18), 392 (99), 130 (32).

 $(\bar{R})$ - $N$ -[3-(1 $\bar{H}$ -Indol-3-yl)-1-thioxo-2-[[(tricyclo-[3.3.1.l<sup>37</sup>]dec-2-yloxy)carbonyl]amino]propyl]-(S)-phenylalanine, Methyl Ester (16). The amide 15 (264 mg, 0.486 mmol) and 2,4-bis(phenylthio)-1,3-dithia-2,4-diphosphetane 2,4-disulfide<sup>12</sup> (201 mg, 0.492 mmol) were stirred for 18 h at room temperature in THF (15 mL). The solvent was removed in vacuo, the crude material was placed in a silica gel column, and elution with dichloromethane followed by a dichloromethane-ethyl acetate gradient (2-8%) gave unreacted amide (50 mg, 19%) and the thioamide 16 (194 mg, 71%):  $[\alpha]^{20}$ <sub>D</sub> = +7.4° (c = 0.97, MeOH);<br>IR (film) 2906, 2855, 1738, 1698, 1499, and 1213 cm<sup>-1</sup>; NMR  $(CDCI_3)$   $\delta$  1.44-2.02 (14 H, m), 2.78 (1 H, m), 2.99 (1 H, m), 3.38 (2 H, m), 3.58 (3 H, s), 4.78 (2 H, br s), 5.25 (1 H, m), 5.55 (1 H, br s), 6.58-6.83 (2 H, m), 7.02 (1 H, s), 7.04-7.25 (5 H, m), 7.33 (1 H, d, *J* = 8 Hz), 7.69 (2 H, m), 8.03 (1 H, s); MS *m/e* (CI) 560  $(100)$   $(M^+ + H)$ .

 $\operatorname{Tricyclo[}3.3.1.1^{3.7}] \text{dec-2-yl}\left[{\bm{R}}{\text{-}}({\bm{R}}^*{\bm{S}}^*)\right] {\text{-}} \left[2{\text{-}}{\text{-}}{\text{}}{\text{-}}{\text{Hydroxy}}\right]$ methyl)-2-phenylethyl]amino]-l-(lff-indol-3-ylmethyl)-2 thioxoethyl]carbamate (17). Lithium aluminum hydride (1.0 M in THF,  $750 \mu L$ , 0.750 mmol) was added dropwise with stirring to a cooled (ice/salt) solution of 16 (104 mg, 0.186 mmol) in THF (5 mL). After 10 min the reaction was quenched via the cautious addition of 2 N HC1 (7.5 mL). The resulting emulsion was extracted with EtOAc  $(3 \times 35 \text{ mL})$ . The organic phase was washed with brine, dried  $(MgSO<sub>4</sub>)$ , filtered, and concentrated in vacuo. The crude material was purified by silica gel column chromatography using dichloromethane-ethyl acetate (65:35) as eluant to give 17 (69 mg, 70%) as a white foam;  $[\alpha]^{20}$ <sub>D</sub> = -42° (c = 0.59, MeOH); IR (film) 2910, 2855,1694,1497, and 1214 cm"<sup>1</sup> ; NMR (CDCI3) « 1.40-2.08 (14 H, m), 2.40 (1 H, m), 2.64 (1 H, m), 3.12-3.64 (5 H, m), 4.58-4.82 (3 H, m), 5.74 (1 H, br s), 7.02 (3 H, m), 7.08-7.27 (5 H, m), 7.35 (1 H, d, *J* = 8 Hz), 7.64 (1 H, br d, *J* = 8 Hz), 7.73 (1 H, d, *J* = 8 Hz), 8.02 (1 H, s); MS *m/e* (CI)  $532$  (2%) (M<sup>+</sup> + H), 498 (100), 346 (40), 130 (20).

 $N$ -Methyl- $N$ -[ $N$ -[(tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yloxy) carbonyl]-D-tryptophyl]-(S)-phenylalanine, Methyl Ester (18). Acid 3 (808 mg, 2.11 mmol), 1-hydroxybenzotriazole monohydrate (348 mg, 2.58 mmol), and N<sub>r</sub>N'-dicyclohexylcarbodiimide (498 mg, 2.41 mmol) were stirred in DMF (2 mL) for 1 h at room temperature, and then  $N$ -methyl- $(S)$ -phenylalanine methyl ester (Bachem) (620 mg, 3.21 mmol) was added. The mixture was stirred for 24 h at room temperature then poured into  $H_2O$  (15 mL). The emulsion was extracted with  $Et_2O$  (30 mL). The organic phase was washed with brine, dried  $(MgSO<sub>4</sub>)$ , and filtered, and the filtrate was concentrated in vacuo. The crude material was purified by silica gel chromatography eluting with ether-hexane (70:30) to give 18 as a white foam (720 mg, 61%);  $[\alpha]^{\mathfrak{D}}_{\mathbb{D}} = -42^{\circ}$  (C = 1.1, CHCl<sub>3</sub>); IR (film) 2909, 2854, 1741, 1699, and 1641 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.42-2.11 (14 H, m), 2.61 (3 H, s), 2.73-2.98 (3 H, m), 3.29 (1 H, m), 3.70 (3 H, s), 4.77 (1 H, s), 4.95 (1 H, m), 5.34 (1 H, m), 5.55 (1 H, d, *J* = 8 Hz), 6.54 (1 H, s), 7.01-7.36 (8 H, in), 7.59 (1 H, d, *J* = 8 Hz), 7.82 (1 H, s); MS *m/e* (CI) 558 (81%) (M\* + H), 130 (100).

 $\bm{{\rm Tr}}$ icyclo $[\![3.3.1.1^{3.7}]\!]$ dec-2-yl  $[\![R\!-\!(R^*,\!S^*)]\!]$ - $[\![2\!-\![[1\!-\!(\mathrm{Hydroxy}$  $methyl-2-phenylethyl[methylamino]-1-(1H-indol-3-y]$ methyl)-2-oxoethyl]carbamate (19). Lithium borohydride (Aldrich) (2.0 M in THF, 0.50 mL, 1.00 mmol) was added dropwise with stirring to a chilled (ice/salt) solution of 18 (510 mg, 0.915 mmol) in THF (5 mL). After the addition was complete, the cooling bath was removed, and the reaction was stirred at room temperature for 18 h. The solution was then cooled, the reaction was quenched with 2 N HG1 (30 mL), and the emulsion was extracted with EtOAc  $(3 \times 30 \text{ mL})$ . The organic phase was washed with brine, dried (MgS04), filtered, and concentrated in vacuo. The crude product was purified by reverse-phase chromatography using methanol-water (4:1) as eluant to give 19 as an amorphous white solid (100 mg, 20%);  $[\alpha]^{23}$ <sub>D</sub> = -16° (c = 0.34, MeOH); IR (film) 3311, 2908, 2855, 1694, and 1631 cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) (a pair of amide rotamers as determined by saturation transfer experiments: irradiation 6f *&* 6.67 (s) resulted in transfer of irradiation to  $\delta$  6.93 (s), indicating that these signals arise from rotamers) *&* 1.44-2.14 (14 H, m), 2.41-3.29 (9.5 H, m), 3.60 (1.5 H, m), 4.61 and 4.87 (1 H, 2 m), 4.78 (1 H, s), 5.40 and 5.62 (1 H, 2 d, *J* = 8 Hz), 6.67 and 6.93 (1 H, 2 s), 6.97-7.48 (8 H, m), 7.59 and 7.81 <1 H, 2 d, *J* = 7 Hz), 8.18 and 8.53 (1 H, 2 s), MS  $m/e$  CI 530 (M<sup>+</sup> + H) (100%).

(±)-a-(Hydroxymethyl)benzenepropanoic Acid (20). Diisopropylamine (28 mL, 0.20 mol) was dissolved in THF (300 mL), cooled to  $-70$  °C, and treated with *n*-butyllithium (1.6 M in hexane, 125 mL, 0.20 mol). After stirring for 20 min at  $0 °C$ , hydrocinnamic acid (14.3 g, 95.2 mmol) was added and the mixture was stirred for a further 15 min at 20 °C. Then formaldehyde gas [generated by heating paraformaldehyde (10.1 g) at 180-220 **°C over 1 h] was passed into the mixture which was kept between 5-15 °C. After a further 10 h, the mixture was acidified with dilute aqueous HC1 and extracted with ether (3 X 100 mL). The organic**  layers were dried (MgSO<sub>4</sub>) and concentrated in vacuo to give a **solid, which was recrystallized from ether to give 20 (9.2 g, 54%); IR (film) 3600-2400 and 1709 cm"<sup>1</sup> ; NMR (DMSO-dg) 8 2.62 (1 H**, m, CHCOOH), 2.78 (2 H, m, PhCH<sub>2</sub>), 3.40 (2 H, m, CH<sub>2</sub>OH), **4.0-3.0 (1H, br, COOH), 7.20 (5 H, m, Ph); MS (FAB)** *m/e* **179 (60) and 127 (100).** 

**(±)-a-[(Acetyloxy)methyl]benzenepropanoic Acid (21). Acetyl chloride (904** *pL,* **12.7 mmol) was added dropwise to a cooled (ice bath) solution of racemic 2-benzyl-3-hydroxypropionic acid (1 equiv) (20) in THF (10 mL). The solution was allowed to warm to room temperature and stirred for 48 h. The solvent was removed in vacuo, and the crude material was purified by silica gel chromatography using toluene-acetic acid (90:10) as eluant to give 21 as a viscous oil (2.04 g, 86%): IR (film) 1742**  and 1714 cm<sup>-1</sup>; NMR (acetone-d<sub>e</sub>)  $\delta$  1.98 (3 H, s), 2.81-2.94 (1 H, **m), 2.96-3.09 (2 H, m), 4.10-4.23 (2 H, m), 7.16-7.32 (5 H, m); MS** (CI)  $m/e$  240 (100%)  $(M^+ + NH_4^+)$ , 223 (42),  $(M^+ + H)$ , 222 **(13) (M<sup>+</sup> ), 204 (4), 162 (2), 91 (4).** 

 $Tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yl [1-(1H-Indol-3-ylmethyl)-1$ **methyl-2-[(2-phenylethyl)amino]ethyl]carbamate (24). To a solution of lithium borohydride (4 mL, 2 M solution, 8 mmol) in THF was added a solution of chbrotrimethylsilane (1.75 g, 16.0 mmol) in THF (5 mL). A white precipitate (of lithium chloride) was observed. After 2 min, a solution of 237a (1 g, 2 mmol) in THF (15 mL) was added slowly (over 3-4-min period), and the reaction mixture was stirred for 20 h at ambient temperature. The reaction**  was treated cautiously with MeOH (5 mL), and the volatiles were **removed in vacuo at 40 °C. The residue was purified by silica gel chromatography using hexane-EtOAc (80:20) as eluant to give 24 as a colorless oil (0.14 g, 14%) and recovered 23 (0.52 g; yield of 24 was based on recovered starting material 30%). The amine 24 (0.14 g, 0.28 mmol) was dissolved in MeOH (5 mL) and treated with 4-toluenesulfonic acid hydrate (0.054 g, 0.28 mmol). The solution was evaporated to leave a white solid (0.19 g, 0.28 mmol):**   $[\alpha]^{20}$ <sub>D</sub> = +22° (c = 0.25, MeOH); IR (film) 2928 and 1708 cm<sup>-1</sup> **(C=0 urethane); NMR (DMSO-dg) 8 1.2 (3 H, s, CH3), 1.4-2.1 (14 H, m, adamantyl), 2.3 (3 H, s, PhCHs), 2.9-3.7 (8 H, m, 4 x CH2), 4.7 (1H, br s, adamantyl H-2), 6.9-7.6 (15 H, m, aromatics, urethane NH), 8.3 (1 H, br, one of**  $NH_2^+$ **), 8.5 (1 H, br, one of**  $PH_2^+$ **), 8.5 (1 H, br, one of NH<sup>2</sup> ), 11.0 (1 H, s, indole NH); MS** *m/e* **(FAB) 486 (100) (M<sup>+</sup> + H), 136 (52).** 

 $Tricyclo[3.3.1.1^{3.7}]dec-2-yl [R-(R^*,S^*)]-[2-[Acetyl(2$ phenylethyl)amino]-1-(1H-indol-3-ylmethyl)-1-methyl**ethyl]carbamate (25). To a solution of 24 (0.1 g, 0.2 mmol) in dichloromethane (20 mL) at 0 °C was added acetyl chloride (0.3 mL, 4 mmol), followed by triethylamine (0.2 mL, 1.4 mmol). Stirring was continued at 0 °C for 20 min, and then the reaction mixture was dissolved in ethyl acetate (50 mL), washed (dilute**  HCl, then H<sub>2</sub>O, and then aqueous NaHCO<sub>3</sub>), dried (MgSO<sub>4</sub>), and **evaporated to dryness. The residue was purified by column chromatography using hexane-ethyl acetate (70:30) as eluant to give an off-white solid (0.092 g, 85%):**  $[\alpha]^{21}$ **D** = +24° (c = 0.25, **CHCI3); IR (film) 2909,2855,1709, and 1610 cm"<sup>1</sup> ; NMR (CDCI3) 8 1.3 (3 H, s), 1.4-2.1 (17 H, m), 2.8 (2 H^m), 3.1 (1 H, d,** *J* **= 14 Hz), 3.3-3.7 (4 H, m), 4.0 (1 H, d,** *J* **= 14 Hz), 4.8 (1 H, s), 5.5 (1 H, br s), 6.9-7.7 (10 H, m), 8.1 (1 H, br s); MS** *m/e* **(FAB) 528 (45) (M<sup>+</sup> + H), 397 (100), 333 (27).** 

**Ethyl** (*R*)-4-[[3-(1*H*-Indol-3-yl)-2-methyl-2-[[(tricyclo-**[3.3.1.l<sup>3</sup> ' 7 ]dec-2-yloxy)carbonyl]amino]propyl](2-phenylethyl)amino]-4-oxobutanoate (26). Prepared by a method similar to that for compound 25. Obtained 26 as a white solid**   $(0.102 \text{ g}, 81\%): [\alpha]^{21}$ **D** = +26° (c = 0.5, CHCl<sub>a</sub>); IR (film) 2905, 2853, 1732, 1711, and 1634 cm<sup>-1</sup>; NMR (CDCl<sub>a</sub>)  $\delta$  1.2 (6 H, m), **1.3-2.1 (14 H, m), 2.6 (4 H, br s), 2.9 (2 H, m), 3.0 (1 H, d,** *J* **= 14 Hz), 3.3-3.8 (4 H, m), 4.0-4.2 (3 H, m), 4.8 (1 H, br s), 5.3 (1 H, br), 6.9-7.6 (10 H, m), 8.1 (1H, br s); MS** *m/e* **(FAB) 614 (36) (M<sup>+</sup> + H), 483 (100), 419 (24).** 

**Ethyl** *(R***)-7-[[3-(l.ff-IndoI-3-yl)-2-methyl-2-[[(tricyclo- [3.3.1.l<sup>37</sup>]dec-2-yloxy)carbonyl]amino]propyl](2-phenylethyl)amino]-7-oxoheptanoate (27). Prepared by a method similar to that for compound 25. (Ethyl pimelyl chloride was prepared from monoethyl pimelate and oxalyl chloride.) Obtained 27** as a white solid  $(0.092 \text{ g}, 68\%)$ :  $[\alpha]^{\omega}$ <sub>D</sub> = +12° (c = 0.25, CHCI<sub>3</sub>);

**IR (film) 2929,2855,1730 (ah), 1712, and 1631 cm"<sup>1</sup> ; NMR (CDC13) 81.2-2.3 (30 H, m), 2.8 (2 H, m), 3.1 (1 H, d, J = 14 Hz), 3.3-3.7 (4 H, m), 4.0-4.2 (3 H, m), 4.8 (1 H, br), 5.6 (1 H, br), 6.9-7.4 (9 H, m), 7.6 (1 H, d,** *J* **= 8 Hz), 8.1 (1 H, br); MS** *m/e* **(FAB) 656 (40) (M<sup>+</sup> + H), 525 (100), 461 (28).** 

 $(R)$ -4-[[3-(1 $H$ -Indol-3-yl)-2-methyl-2-[[(tricyclo-**[3.3.1.l<sup>8</sup> ' 7 ]dec-2-yloxy)carbonyl]amino]propyl](2-phenylethyl)amino]-4-oxobutanoic Acid (28). To a stirred solution of 26 (0.04 g, 0.06 mmol) in THF (5 mL) were added methanol (5 mL), water (5 mL), and lithium hydroxide monohydrate (0.1 g, 2.4 mmol). The reaction mixture was stirred for 40 min at**  ambient temperature and then acidified (2 N HCl, 50 mL), and **the products were extracted into ethyl acetate (50 mL). The organic phase was dried (MgS04) and evaporated in vacuo (40 °C). The oily residue was purified by column chromatography using dichloromethane/methanol (9:1) as eluant to give 28 (0.031 g, 81%) as a white solid: IR (film) 3400 (br), 2912, 2852,1714, 1700, and 1635 cm"<sup>1</sup> ; NMR (MeOH-d4) 8 1.1-2.2 (17 H, m, CH3> adamantane**), 2.6–3.2 (6 H, m, 3  $\times$  CH<sub>2</sub>), 3.4–4.1 (6 H, m, 3  $\times$  CH<sub>2</sub>),  $4.8$  (1 H, br, adamantane H-2), 7.0–7.6 (10 H, m, aromatics);  $[\alpha]^{\vec{2}}$  $= +4^{\circ}$  (c = 0.2, MeOH); MS  $m/e$  (FAB) 586 (51) (M<sup>+</sup> + H), 455 **(100), 391 (24).** 

 $Tricyclo[3.3.1.1<sup>3,7</sup>]$ dec-2-yl  $(R)$ -[1-(1*H*-Indol-3-yl**methyl)-2-methyl-2-[(2-phenylethyl)amino]-2-thioxoethyl] carbamate (29). To a solution of 237a (0.1 g, 0.2 mmol) in toluene (10 mL) was added Lawesson's reagent (0.10 g, 0.25 mmol), and the mixture was heated to reflux for 1 h. The mixture was allowed to cool to ambient temperature and was purified by silica gel chromatography using CH2C12 followed by ether as eluant to give 29 as a white foam (0.065 g, 63%):**  $[\alpha]^{\mathbb{Z}}_D = +40^{\circ}$  (c = 0.3, CHCI<sub>3</sub>); **IR** (film) 2916, 1703 (C=0 urethane), and 1520 (C-S) cm<sup>-1</sup>; NMR **(CDCI3) 8 1.5 (3 H, s, CH3), 1.6-2.0 (14 H, m, adamantane), 2.6 (2 H, m, CH2Ph), 3.4 (1 H, d,** *J* **= 14 Hz, one of CH2-indole), 3.6**   $(1 H, d, J = 14 Hz$ , one of  $CH_2$ -indole), 3.8  $(2 H, m, CH_2N)$ , 4.7 **(1 H, br, adamantane H-2), 5.3 (1 H, br, urethane NH), 6.9-7.5 (9 H, m, indole H-5, H-6, H-7, H-2, phenyl), 7.6 (1 H, d,** *J* **= 8 Hz, indole H-4), 7.8 (1 H, br, thioamide NH), 8.1 (1 H, br, indole NH); MS** *m/e* **(FAB) 516 (25) (M<sup>+</sup> + H), 217 (95), 109 (100).** 

**a-Methyl-JV-[(tricyclo[3.3.1.1S7]dec-2-yloxy)carbonyl]-** (R)-tryptophan, 2-phenylethyl ester (30) was prepared and **purified by a method similar to that for compound 36: yield 103 mg** (85%);  $[\alpha]^{\mathbb{Z}}_D = +13.6^{\circ}$  (c = 1, MeOH); IR (film) 3412 (NH), **2907, 2855, 1730 (C=0 ester), and 1698 (C=0 urethane) cm<sup>-1</sup>; NMR (CDCI3) 81.50 (1H, s, adamantyl CH), 1.56 (4 H, s, CCH3, adamantyl CH), 1.70-2.05 (12 H, m, adamantyl CH), 2.87 (2 H,**   $J = 7$  Hz, CH<sub>2</sub>CH<sub>2</sub>Ph), 3.32 (1 H, d,  $J = 14.5$  Hz, one of **CHrindole), 3.45-3.55 (1H, m, one of CH2-indole), 4.18 (1H, dt,**   $J = 7, 11$  Hz, one of OCH<sub>2</sub>), 4.28–4.38 (1 H, m, one of OCH<sub>2</sub>), 4.84 **(1 H, s, CHOCONH), 5.3-5.5 (1 H, br s, OCONtf), 6.79 (1H, s, indole H-2), 7.05 (1 H, t,** *J* **= 7 Hz), 7.10-7.33 (7 H, m), 7.47 (1 H, d,** *J* **- 8 Hz, indole H-4), 8.00 (1 H, s, indole NH); MS** *m/e*  **(FAB) 501 (30) (M<sup>+</sup> + H), 217 (84), 130 (100).** 

**Tricyclo[3.3.1.1\*']dec-3-yl (fl)-[l-(lIT-indol-3-ylmethyl)-l-methyl-2-oxo-2-[(2-phenylethyl)thio]ethyl]carbamate (31) was prepared and purified by a method similar to that for compound 36: yield 196 mg (57%);**  $[\alpha]^{\{2\}}_{D} = +47^{\circ}$  (c = **0.5, MeOH); IR (film) 3500-3300,2906,2854, and 1698 cm"<sup>1</sup> ; NMR (CDCI3) 8 1.50 (1 H, s, adamantyl CH), 1.53 (4 H, s,** *CCHS,* **ad**amantyl), 1.70-2.05 (12 H, m, adamantyl), 2.84 (2 H, t,  $J = 8$  Hz,  $CH_2Ph$ , 3.08-3.14 (2 H, m,  $CH_2CH_2Ph$ ), 3.30 (1 H, d,  $J = 14$  Hz, one of  $CH_2$ -indole), 3.56 (1 H, d,  $J = 14$  Hz, one of  $CH_2$ -indole), **4.87 (1 H, s, CffOCONH), 5.00-5.20 (1 H, br s, OCONH), 6.97 (1H, d,** *J* **= 2 Hz, indole CH), 7.10 (1 H, t,** *J* **= 7 Hz, indole CH), 7.15-7.32 (6 H, m, Ph, indole CH), 7.35 (lH,d,J= 8 Hz, indole CH), 7.59 (1H, d,** *J* **= 8 Hz, indole CH), 8.10 (1 H, s, indole NH); MS** *m/e* **(FAB) 517 (11) (M<sup>+</sup> + H), 379 (55), 351 (39), 307 (100).** 

 $\bf Tricyclo[3.3.1.1^{3.7}] dec-2-yl$  ( $\bf R$ )-[1-(1 $\bf H$ -indol-3-yl**methyl)-l-methyl-2-[methyl(2-phenylethyl)amino]-2-oxoethyl]carbamate (32) was prepared and purified by a method**  similar to that for compound 36: yield 50 mg (61%);  $[\alpha]^2$ <sup>D</sup> = **+67.2° (c = 0.5, MeOH); IR (film) 3500-3200, 2907, 2855,1698 (0= 0 urethane), and 1625 (0= 0 amide) cm"<sup>1</sup> ; NMR (DMSO-d6) (340 K) 8 1.30 (3 H, s, CCH3), 1.40-1.55 (2 H, m, adamantyl), 1.65-2.05 (12 H, m, adamantyl), 2.70-2.80 (2 H, m,** *CH2Ph),* **2.99 (3 H, s, NCH3), 3.20 (1 H, d,** *J =* **15 Hz, one of CH2-indole), 3.44 (1 H, d,** *J* **= 15 Hz, one of CH2-indole), 3.40-3.70 (2 H, m,** 

#### *Amide Replacements in CCK-B "Dipeptoids"*

 $CH_2CH_2NHCO$ , 4.74 (1 H, s, CHOCONH), 6.90-7.05 (4 H, m), 7.15-7.30 (5 H, m), 7.31 (1 H, d, *J* = 8 Hz, indole 7-H), 7.43 (1 H, d,  $J = 8$  Hz, indole 4-H), 10.65 (1 H, s, indole NH); MS  $m/e$  $(FAB) 514 (66) (M<sup>+</sup> + H), 379 (76), 318 (32), 173 (34), 135 (100).$ **Methyl** *(R* **)-4-[ (2-Hydroxy- 1 -phenylethyl)amino]-4-oxobutanoate** (33). Prepared by a route similar to that for compound **35.** Recovered 3.6 g (65%) of 33:  $[\alpha]^{22}$ <sub>D</sub> = -53° (c = 1, CHCl<sub>3</sub>); IR (film) 3250 (OH), 1737 (C=0 ester), and 1651 (C=0 amide) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.7 (1 H, br, OH), 2.5-2.8 (6 H, m, 3 × CH<sub>2</sub>), 3.7 (2 H, s, CH3), 5.1 (1 H, m, CH), 6.4 (1 H, br d, NH), 7.2-7.4 (5 **H,** m, Ph); MS m/e **(FAB)** 252 (100) **(M<sup>+</sup> + H),** 225 (23), 115  $(26)$ 

 $Tricyclo[3.3.1.1<sup>3,7</sup>]$ dec-2-yl  $3-(1H-Indol-3-ylmethyl)-3$ **methyl-4,9-dioxo-7-phenyl-5,13-dioxa-2,8-diazatetradecanoate (34).** To a solution of 1,3-dicyclohexylcarbodiimide (0.3 g, 1.5 mmol) and 4-(dimethylamino)pyridine (0.05 g, 0.40 mmol) in CH2C12 (40 mL) was added **22** (0.50 g, 1.3 mmol). After stirring for 20 min at ambient temperature, **33** (0.30 g, 1.2 mmol) was added and the reaction was heated at reflux for 2 h. The volatiles were removed in vacuo at 40 °C, and the residue was purified by silica gel chromatography, using hexane-EtOAc as eluant to give 34 as a white solid (0.41 g, 52%):  $[\alpha]^{22}$ <sub>D</sub> = -18° (c = 1, CHCl<sub>3</sub>); IR (film) 2920, 1739 (C=0 ester), 1700 (C=0 urethane), and 1660 (C=O amide) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.5-2.1 (17 H, m, adamantane, CH<sub>3</sub>), 2.5-2.7 (4 H, m, 2  $\times$  CH<sub>2</sub>), 3.4 (2 H, m, CH<sub>2</sub>-indole), 3.7 (3 H, s, CH<sub>3</sub>), 4.1 (1 H, m, one of CH<sub>2</sub>O), 4.8 (2 H, br, adamantane H-2, one of  $CH<sub>2</sub>O$ ), 5.2 (1 H, br s, urethane NH), 5.3 (1 H, m, OH), 6.9-7.4 (10 H, m, indole H-5, H-6, H-7, H-2, amide NH, phenyl), 7.5 (1H, d, *J* = 8 Hz, indole H-4), 8.3 (1H, br, indole NH); MS  $m/e$  (FAB) 631 (11) (M<sup>+</sup> + H), 307 (10), 252 (12), 234  $(100)$ 

**Methyl** *(R***)-4-[(2-Hydroxy-l-phenylethyl)amino]-4-oxo-2-butenoate (35).** To a solution of monomethyl fumarate (3.0 g, 23 mmol) in EtOAc (40 mL) was added 1-hydroxybenzotriazole hydrate (3.4 g, 22 mmol), followed by  $N,N$ -dicyclohexylcarbodiimide (4.5 g, 22 mmol), and the reaction was stirred at ambient temperature for 1 h. The solid was filtered off and discarded. To the filtrate was added  $(R)$ - $\alpha$ -phenylglycinol (3.0 g, 22 mmol), and stirring was continued for 20 min. The volatiles were removed in vacuo at 40 °C, and the residue was purified by silica gel chromatography using hexane-EtOAc (1:1) as eluant to give the amide 35 as a white solid  $(2.5 \text{ g}, 46\%)$ :  $[\alpha]^{24}$ <sub>D</sub> = -53° (c = 1,  $CHCl<sub>3</sub>$ ); IR (film) 3250 (OH), 1729 (C=O ester), 1666 (C=O amide), and 1640 (C=C) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  2.3 (1 H, t, J = 5 Hz, OH), 3.8 (3 H, s, CH<sub>3</sub>), 3.9 (2 H, t,  $J = 3$  Hz, CH<sub>2</sub>), 5.2 (1) H, m, CH), 6.6 (1 H, br d, NH), 6.8 (1 H, d, *J* = 15 Hz, trans alkene), 7.0 (1 H, d, *J* = 15 Hz, trans alkene), 7.3-7.4 (5 **H,** m, Ph); MS m/e (FAB) 250 (100) **(M<sup>+</sup>** + 1).

 $Tricyclo[3.3.1.1<sup>3,7</sup>]$ dec-2-yl  $3-(1H-Indol-3-ylmethyl)-3$ **methyl-4,9-dioxo-7-phenyl-5,13-dioxa-2,8-diazatetradec-10 enoate (36).** To a solution of  $N$ , $N'$ -carbonyldiimidazole (0.15 g,  $(0.90 \text{ mmol})$  in  $\text{CH}_2\text{Cl}_2$  (40 mL) was added compound  $22^1$  (0.25) g, 0.63 mmol). After 20 min of stirring at ambient temperature, **35** (0.2 g, 0.8 mmol) was added and the reaction was heated at reflux for 10 h. After cooling to ambient temperature, the volatiles were removed in vacuo at  $40^{\circ}$ C, and the residue was purified by silica gel chromatography using hexane-EtOAc (70:30) as eluant to give 36 as a white solid (0.28 g, 71%):  $[\alpha]^{24}$ <sub>D</sub> = +25° (c = 0.2, CHCl<sub>3</sub>); IR (film) 2910, 1730 (C=O ester), 1695 (C=O urethane), and 1670 (C=0 amide) cm"<sup>1</sup> ; NMR (CDC13) *&* 1.4-2.1 (18 H, m, adamantane, CH<sub>3</sub>), 3.3 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 3.5 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 3.8 (3 H, s, CH<sub>3</sub>), 4.1  $(1 H, dd, J = 4, 11 Hz, one of CH<sub>2</sub>O), 4.8 (1 H, br s, adamantane)$ H-2), 5.0 (1 H, br, CH), 5.1 (1 H, s, urethane NH), 5.3 (1 H, br, one of CH20), 6.8 (1 H, d, *J* = 16 Hz, trans alkene CH), 6.9-7.3 (10 H, m, indole H-6, H-7, H-2, phenyl, amide NH, one of alkene CH), 7.4 (1 H, d, *J =* 8 Hz, indole H-5), 7.5 (1 H, d, *J* = 8 Hz, indole H-4), 8.3 (1 H, br, indole NH); MS *m/e* (FAB) 628 (27) (M<sup>+</sup> - **H),** 232 (38), 130 (100).

 $\bf Tricyclo[3.3.1.1<sup>3.7</sup>]$ dec-2-yl $(R)$ -[2-Hydroxy-1- $(1H\text{-}\mathrm{indol-3})$ **ylmethyl)-l-methylethyl]carbamate (38).** To a solution of **37<sup>1</sup>**  $(1.0 \text{ g}, 2.4 \text{ mmol})$  in THF  $(20 \text{ mL})$  at  $0^{\circ}$ C was added a solution of lithium aluminum hydride in ether (3 mL of a 1 M solution, 3 mmol), and the reaction was stirred for 20 min. EtOAc (20 mL) was added slowly, and the resulting solution was washed with acid  $(2 \text{ N } HCl, 2 \times 100 \text{ mL})$ , dried  $(MgSO<sub>4</sub>)$ , and evaporated to dryness

to give 38 as a white solid  $(0.85 \text{ g}, 91 \text{ %})$ :  $[\alpha]^{22}$ <sub>D</sub> = +42<sup>o</sup> (c = 1,  $CHCl<sub>3</sub>$ ; IR (film) 2918 and 1693 (C=0 urethane) cm<sup>-1</sup>; NMR (CDCI3) *6* 1.2 (3 H, s, CH3), 1.5-2.1 (14 H, m, adamantane), 3.0 (1H, d, *J =* 14 Hz, one of CH2-indole), 3.25 (1 H, d, *J* = 14 Hz, one of CH<sub>2</sub>-indole), 3.8 (2 H, m, CH<sub>2</sub>O), 4.0 (1 H, br, OH), 4.8 (1H, br, adamantane H-2), 4.85 (1 H, br s, urethane NH), 7.0-7.2 (3 H, m, indole H-6, H-7, H-2), 7.35 (1H, d, *J* = 8 Hz, indole H-5), 7.6 (1 H, d, *J* = 8 Hz, indole H-4), 8.1 (1 H, br, indole NH); MS *m/e* (FAB) 383 (50) (M<sup>+</sup> + H) 252 (45) 135 (100).

**Tricyclo[3.3.1.1<sup>87</sup>]dec-2-yl(fl)-[l-Formyl-2-(lff-indol-3 yl)-l-methylethyl]carbamate (39).** To a solution of **38** (0.03 g, 0.08 mmol) in  $CH_2Cl_2$  (40 mL) at ambient temperature was added N-methylmorpholine N-oxide (0.1 g, 0.9 mmol), molecular sieves (4A activated powder, 0.5 g) and tetra-n-propylammonium perruthenate (TPAP) (0.01 g, 0.03 mmol). After stirring for 30 min, the solid was removed by filtration, and then the volatiles were removed in vacuo (40 °C). The residue was dissolved in EtOAc and purified by silica gel chromatography using hexane-EtOAc (95:5) as eluant to give **39** as a white foam (0.2 g, 67%):  $[\alpha]^{22}$ <sub>D</sub> = +22° (c = 1, CHCl<sub>3</sub>); IR (film) 2900, 1732 (C=0 aldehyde), and 1692 (0= 0 urethane) cm"<sup>1</sup> ; NMR (CDC13) *d* 1.3-2.1  $(17 H, m, adamantane, CH<sub>3</sub>), 3.3 (2 H, br, CH<sub>2</sub>-indole), 4.8 (1 H,$ s, adamantane H-2), 5.2 (1 H, br, urethane NH), 6.9-7.6 (5 H, m, indole), 8.1 (1 H, br s, indole NH), 9.6 (1 **H,** s, CHO); MS m/e (FAB) 381 (26) (M + **H),** 363 (45), 264 (100), 229 (74).

 $Tricyclo[3.3.1.1<sup>3,7</sup>]dec-2-yl [R-(Z)]-[1-(1H-Indol-3-yl$ **methyl)-l-methyl-5-phenyl-2-pentenyl]carbamate** (40). A mixture of triphenylphosphine (0.35 g, 1.3 mmol) and 1-bromo-3-phenylpropane (0.27 g, 1.3 mmol) was heated to 110 °C at which point the molten reactants solidified. On cooling to ambient temperature and triturating with hexane a white solid was removed (0.4 g, 65%). This was added to a suspension of sodium hydride (50 mg of 50% in oil dispersion, 1 mmol) in toluene (40 mL), and the reaction was heated to reflux for 20 min. The aldehyde **39** (0.2 g, 0.5 mmol) was added, and heating was continued for 1 h. The volatiles were removed in vacuo, and the residue was purified by silica gel chromatography using EtOAc as eluant to give 40 as a white solid  $(0.20 \text{ g}, 79\%): [\alpha]^{23}$ <sub>D</sub> = +1.5°  $(c = 1, CHCl<sub>3</sub>)$ ; IR (neat) 2904, 1696 (C= $O$  urethane), 1683 (C= $C$ ) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  1.4 (3 H, s, CH<sub>3</sub>), 1.4-2.1 (14 H, m, adamantane), 2.4-2.6 (4 H, m,  $2 \times CH_2$ ), 3.1 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 3.3 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 4.8 (2 H, m, urethane NH, adamantane H-2), 5.4 (1 H, dt, *J* = 5,12  $\text{Hz}$ , CH=CH–CH<sub>2</sub>), 5.6 (1 H, d, J = 12 Hz, CH=CHCH<sub>2</sub>), 6.9 (1H, s, indole H-2), 7.0-7.3 (8 H, m, indole H-5, H-6, H-7, phenyl), 7.6 (1 H, d, *J* = 8 Hz, indole H-4), 8.3 (1 H, s, indole NH); MS *m/e* (FAB) 483 (5) (M<sup>+</sup> ), 352 (99), 288 (99), 135 (100).

 $\text{Tricyclo[}3.3.1.1^{3.7}]$ dec-2-yl (*R*)-[1-(1*H*-Indol-3-yl**methyl)-l-methyl-5-phenylpentyl]carbamate (41).** A solution of 40 (0.3 g, 0.6 mmol) in EtOH (20 mL) was treated with 10% Pd/C (0.01 g, 1 mmol) and hydrogen gas at 40 psi for 16 h. After removal of the catalyst by filtration through Kieselguhr, the solvent was removed in vacuo and the residue was purified by silica gel chromatography using hexane-ether (90:10) as eluant to give 41 as a white glassy solid (0.25 g, 83%):  $[\alpha]^{22}$ <sub>D</sub> = +30°  $(c = 0.5, CHCl<sub>3</sub>)$ ; IR (film) 2916 and 1698 (C=0 urethane) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.2 (3 H, s, CH<sub>3</sub>), 1.4-2.1 (20 H, m, adamantane,  $3 \times CH_2$ ), 2.6 (2 H, t,  $J = 8$  Hz, CH<sub>2</sub>Ph), 3.0 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 3.2 (1 H, d,  $J = 14$  H, one of CH<sub>2</sub>-indole), 4.5 (1 H, br, urethane NH), 4.8 (1 H, br, adamantane H-2), 6.9 (1 H, d, *J* = 2 Hz, indole H-2), 7.0-7.3 (8 H, m, indole H-5, H-6, H-7, phenyl), 7.6 (1 H, d, *J* = 8 Hz, indole H-4), 8.2 (1 H, br, indole NH); MS *m/e* (FAB) 485 (3) (M<sup>+</sup>  **+** H), 354 (38), 310 (45), 290 (58), 135 (100).

**Tricyclo[3.3.1.1<sup>8</sup> ' 7 ]dec-2-yl[2-Hydroxy-l-(lff-indol-3-ylmethyI)-l-methyI-5-phenyIpentyl]carbamate (42,**43). To a stirred mixture of magnesium turnings (0.5 g, 21 mmol) and dry ether (20 mL) at 0 °C were addded l-bromo-3-phenylpropane (0.2 mL, 1.3 mmol) and iodine (1 crystal). After a short induction period, a vigorous exothermic reaction was observed. When this reaction had subsided, the supernatant was removed by syringe and was added to a solution of 39 (0.3 g, 0.8 mmol) in ether (40 mL) at 0 °C. After 20 min, the reaction was warmed to ambient temperature and quenched (2 N HC1, 50 mL), and the products were extracted into  $EtOAc$  (2  $\times$  50 mL). The combined extracts were dried  $(MgSO_4)$  and evaporated to dryness. The residue was

purified by silica gel chromatography using hexane-ether (70:30) as eluant to give 42 and 43. Diastereoisomer I as a white solid  $(0.15 \text{ g}, 38\%)$ :  $[\alpha]^{\mathbb{Z}_{D}} = +69^{\circ}$  (c = 0.5, CHCl<sub>3</sub>); IR (film) 2910 and 1687 (C=O urethane) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.3 (4 H, m, 2 × CH<sub>2</sub>), 1.4-2.1 (17 H, m, adamantane, CH<sub>3</sub>), 2.7 (2 H, m, CH<sub>2</sub>Ph), 3.1  $(1 H, d, J = 14 Hz$ , one of  $CH_2$ -indole), 3.5 (1 H, d,  $J = 14 Hz$ , one of CH2-indole), 3.6 (1 H, m, CH), 4.6 (1 H, br, OH), 4.7 (1 H, s, urethane NH), 4.8 (1 H, s, adamantane H-2), 7.0-7.4 (9 H, m, indole H-5, H-6, H-7, H-2, phenyl), 7.6 (1H, d, *J =* 8 Hz, indole H-4), 8.2 (1 H, s, indole NH), MS *m/e* (FAB) 501 (17) (M<sup>+</sup> + H), 135 (100), 130 (99). Diastereoisomer II as a white solid (0.18 g, 46%):  $[\alpha]^{22}$ <sub>D</sub> = +43° (22 °C, c = 0.5, CHCl<sub>3</sub>); IR (film) 2912 and 1690 (C=O urethane) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>)  $\delta$  1.3 (4 H, m, 2  $\times$  CH<sub>2</sub>), 1.5-2.1 (17 H, m, adamantane, CH<sub>3</sub>), 2.7 (2 H, m, CH<sub>2</sub>Ph), 2.9 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 3.2 (1 H, d,  $J = 14$  Hz, one of CH2-indole), 3.8 (1 H, m, CH), 4.0 (1 H, br, OH), 4.7 (1 H, s, urethane NH), 4.8 (1 H, s, adamantane H-2), 6.9-7.4 (9 H, m, indole H-5, H-6, H-7, H-2, phenyl), 7.6 (1H, d, *J =* 8 Hz, indole H-4), 8.1 (1 H, br, indole NH); MS *m/e* (FAB) 501 (12) (M<sup>+</sup> + H), 370 (30), 135 (100), 130 (99).

 $Tricyclo[3.3.1.1^{3.7}]$ dec-2-yl $[R-(R^*,S^*)]$ -[1-[4,5-Dihydro-4-(phenylmethyl)-2-thiazolyl]-2-(lff-indol-3-yl)-l-methyl-

ethyl]carbamate  $(45)$ . To a solution of  $44^{7a}$   $(0.1 g, 0.2 mmol)$ in toluene (10 mL) was added Lawesson's reagent (0.10 g, 0.25 mmol), and the mixture was heated at reflux for 1 h. The reaction mixture was allowed to cool to ambient temperature and was purified by silica gel chromatography using  $CH_2Cl_2$ -ether as eluant to give 45 as a white solid  $(0.07 \text{ g}, 70\%)$ :  $[\alpha]^{\overline{22}}_{\text{D}} = -20^{\circ}$  (c = 0.5,  $CHCl<sub>3</sub>$ ; IR (film) 2910, 1697 (C=0 urethane), and 1620 (C=N) cm<sup>-1</sup>; NMR (CDCl<sub>3</sub>) δ 1.3 (3 H, s, CH<sub>3</sub>), 1.4-2.1 (14 H, m, adamantane),  $2.2$  (1H, br, one of CH<sub>2</sub>Ph),  $2.8$  (1H, br, one of CH<sub>2</sub>Ph), 2.9 (1 H, m, one of CH<sub>2</sub>S), 3.2 (1 H, m, one of CH<sub>2</sub>S), 3.4 (1 H, d,  $J = 14$  Hz, one of CH<sub>2</sub>-indole), 3.7 (1 H, d,  $J = 14$  Hz, one of CH2-indole), 4.6 (1 H, m, CH), 4.9 (1 H, br, adamantane H-2), 5.8 (1 H, br, urethane NH), 6.9-7.4 (9 H, m, indole H-5, H-6, H-7, H-2, phenyl), 7.7 (1H, d, *J* = 8 Hz, indole H-4), 8.1 (1H, br, indole NH); MS  $m/e$  (FAB) 529 (47) (M<sup>+</sup> + H), 398 (44), 130 (100).

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# *Communications to the Editor*

### **Novel Naphthalenic Ligands with High Affinity for the Melatonin Receptor**

#### **Introduction**

The neurohormone melatonin (5-methoxy-N-acetyltryptamine), which is synthesized principally in the pineal gland, is putatively involved in several physiological axes. These include the entrainment of both seasonal (reproduction)<sup>1</sup> and circadian (activity)<sup>2</sup> rhythms. The role of melatonin in the human is still controversial, but it is thought to be involved in the regulation of sleep, $34$  seasonal disorders,<sup>4</sup> depression,<sup>4</sup> and ageing.<sup>5</sup> The lack of rigorous data in this regard has often led to conflicting and confusing reports.<sup>4</sup> Nevertheless, the localization of 2-[<sup>125</sup>I] iodomelatonin binding sites in the suprachiasmatic nucleus  $(SCN)$  of the human hypothalamus,<sup>6</sup> which is the biological clock of the brain, enhances the argument for a physiological role in man.

The recent development of a radioligand binding assay for melatonin receptors using ovine pars tuberalis membranes of the pituitary enables the rapid screening of melatonin analogues for their receptor binding potency.<sup>7</sup>

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Table I. Structure Binding Characteristics of 2-(7-Methoxy-l-naphthyl)ethylamides for the Melatonin Receptor

compd	R	mp, °C	$K_{\sf d}$ :" $mean \pm SEM$	order of potency
			CH2CH2NHCOR	
	$_{\mathsf{H}_3$ CO			
5	CH <sub>3</sub>	109-110	$(1.00 \pm 0.353) \times 10^{-10}$	7
6	CH <sub>2</sub> CH <sub>3</sub>	103	$(2.19 \pm 0.897) \times 10^{-11}$	5
7	$(CH_2)_2CH_3$	96-97	$(6.15 \pm 0.178) \times 10^{-12}$	4
8	$(CH2)3CH3$	90	$(3.43 \pm 2.430) \times 10^{-12}$	3
9	$(CH2)4CH3$	$84 - 85$	$(2.83 \pm 0.211) \times 10^{-7}$	12
10	$\overline{\text{CH}_2\text{)}_5\text{CH}_3}$	68-70	$(2.33 \pm 0.815) \times 10^{-6}$	15
11	isopropyl	$77 - 78$	$(2.29 \pm 0.745) \times 10^{-9}$	8
12	$CH=CHCH3$	119-120	$(8.47 \pm 0.793) \times 10^{-9}$	9
13	cyclopropyl	$91 - 92$	$(4.18 \pm 0.870) \times 10^{-13}$	$\mathbf{1}$
14	cyclobutyl	$75 - 76$	$(2.42 \pm 0.536) \times 10^{-8}$	10
15	cyclohexyl	$105 - 106$	$(1.59 \pm 0.792) \times 10^{-7}$	11
16	benzyl	101–102	$(1.95 \pm 0.397) \times 10^{-6}$	14
17	phenyl	128–130	$(4.99 \pm 4.360) \times 10^{-6}$	16
18	$C_6H_3Cl_2(m)$	138	$(1.13 \pm 0.856) \times 10^{-5}$	17
19	2-indolyl	198–199	$(1.17 \pm 0.781) \times 10^{-6}$	13
	$_{\rm H_3CC}$		CH2CH2NHCOR	
		н		
melatonin	CH <sub>3</sub>	$116 - 118$	$(9.15 \pm 3.980) \times 10^{-11}$	6
20	cyclopropyl	101-102	$(4.97 \pm 0.928) \times 10^{-13}$	$\overline{2}$

 $K_d$ 's are the means of three independent experiments, calculated as inverse *K,'s* from single-site fit of untransformed data to equations describing the law of mass action,<sup>10</sup> using a  $K_a$  for 2- $[1^{25}]$  iodomelatonin of  $3.3 \times 10^{10}$  L/mol. The protocol for the synthesis and purification of the radioligand 2-[<sup>125</sup>I]iodomelatonin and the receptor-binding assay using ovine pars tuberalis membranes were as described previously.<sup>7</sup>

Fifteen naphthalenic bioisosteres of melatonin with variations on the  $N$ -acylamino group have been synthesized.

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